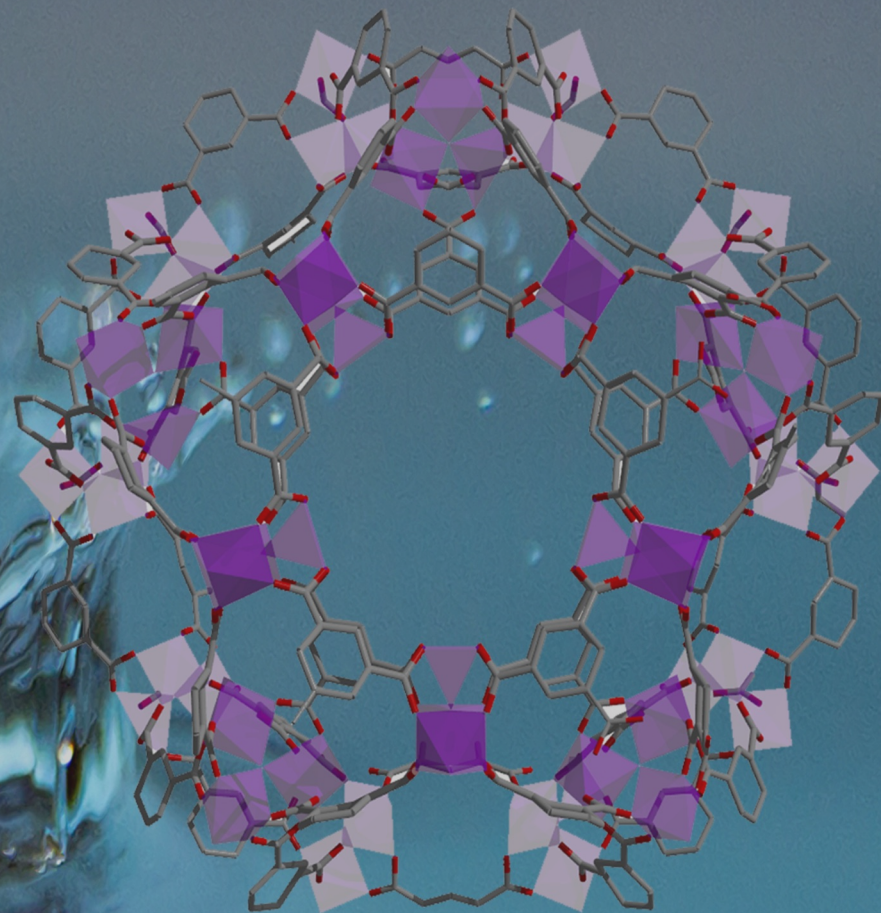


Tuning MOF structures to achieve high performance separations

Clean water in seconds

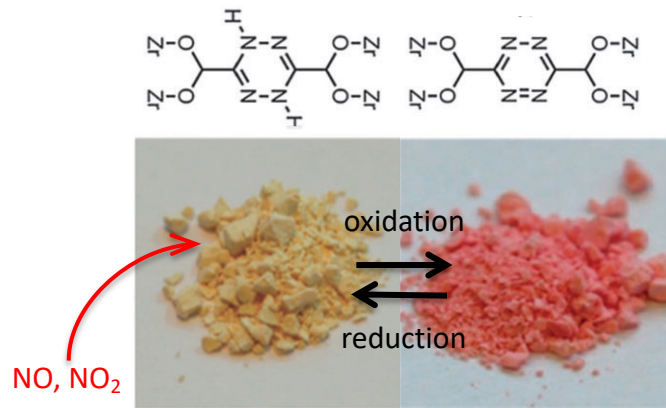


Four primary applications

Storage/Delivery



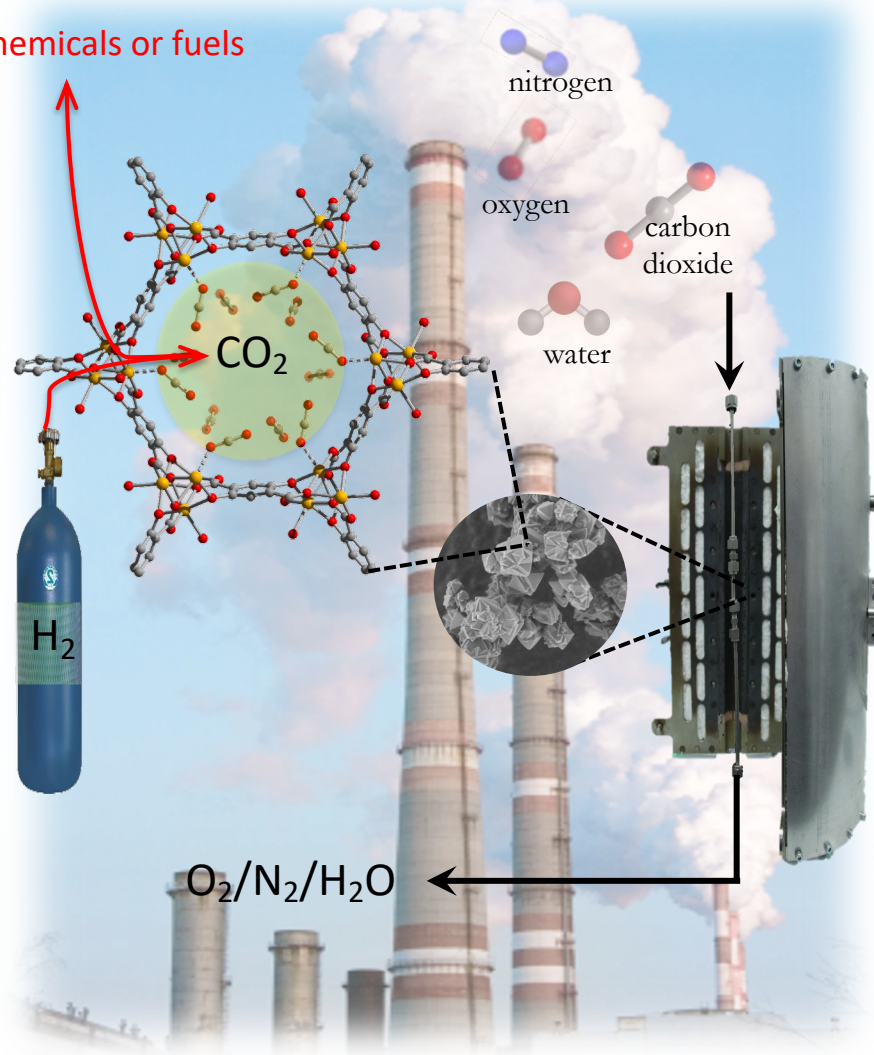
Sensing



Nickerl, Senkovska, Kaskel *Chem. Commun.*, **2015**, 51, 2280.

Separation and conversion

Chemicals or fuels



How much of the world's energy is spent on separations?

15%

Seven separations that can change the world

1. Greenhouse gases from dilute emissions
2. Trace contaminants from water
3. Rare-metals from ore
4. Uranium from sea water



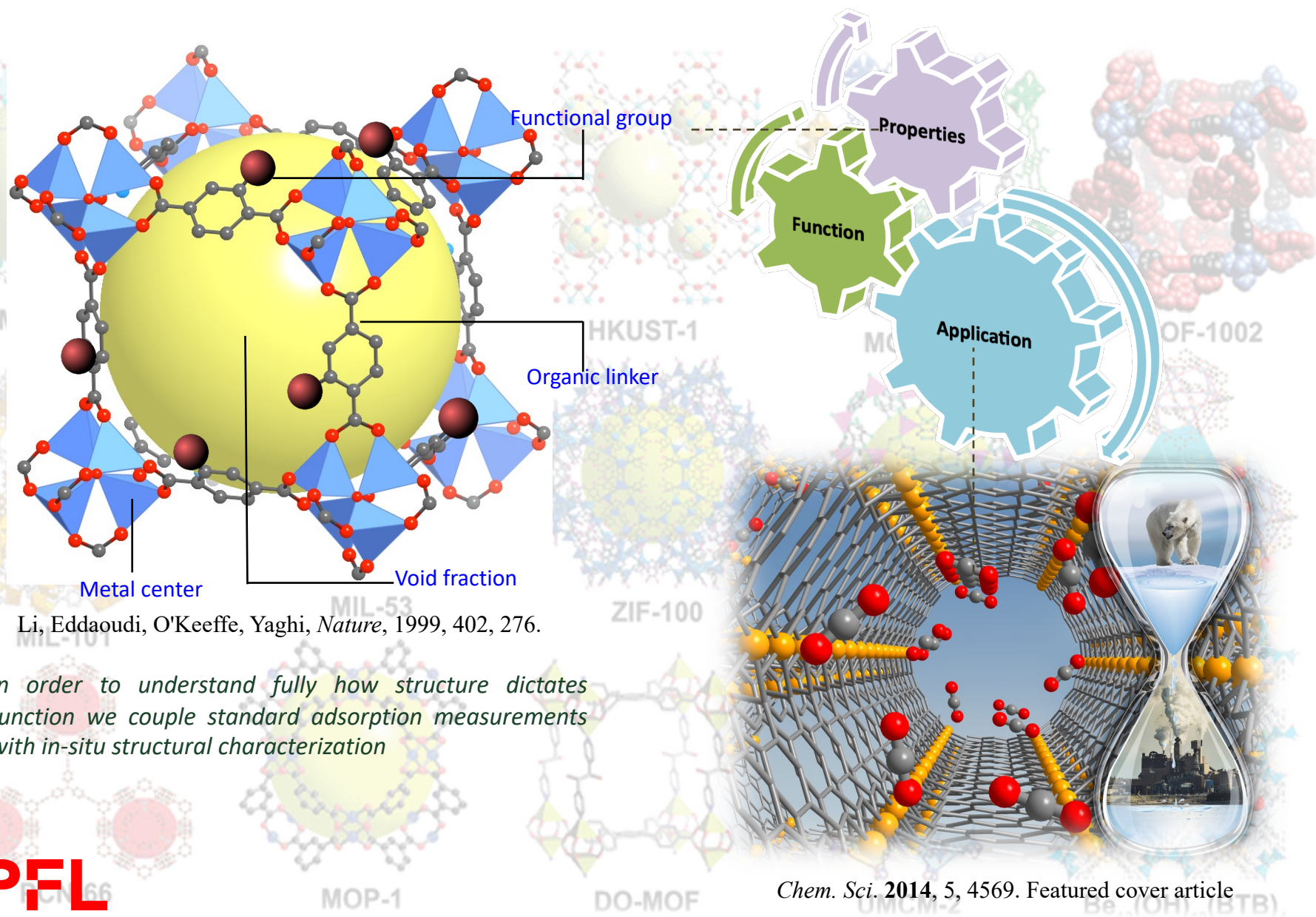
5. Alkenes from alkanes
6. Benzene derivatives
7. Hydrocarbon separation from crude oil

Seven separations that can change the world



- reduce energy input into separations
- make impossible separations possible

Understanding structure-derived function





Climate Action: Elimination of carbon dioxide

Global CO₂ Emissions - 30.3 Gtons/year



Automobiles

22%

6.4 Gtons/year

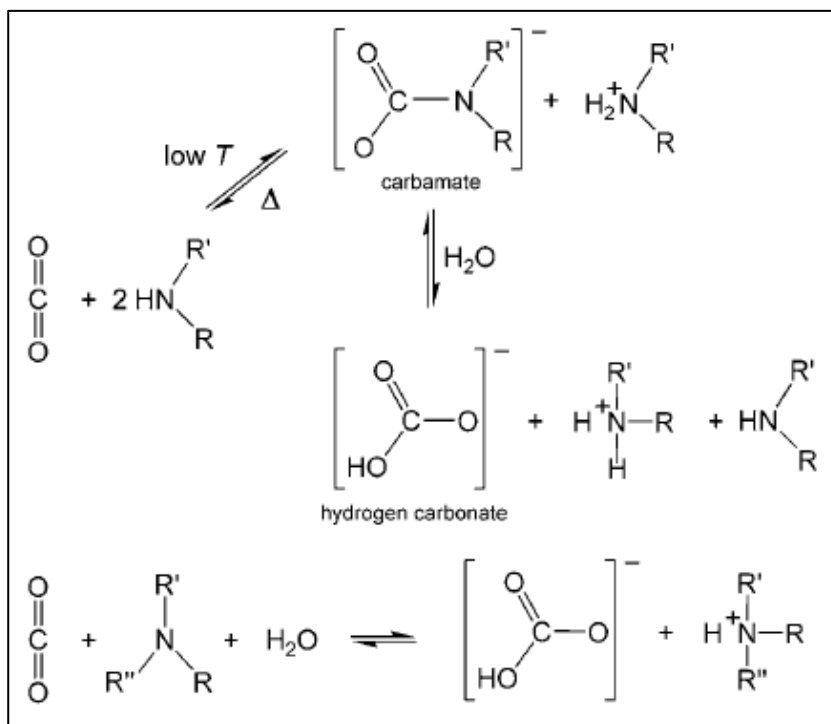


electricity and heat

41%

12.4 Gtons/year

Most mature CO₂ capture technology



30% Parasitic Energy Cost

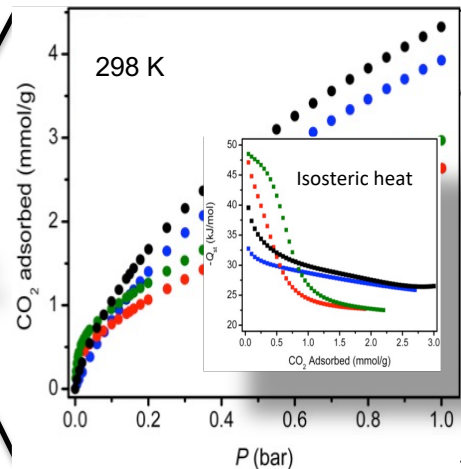
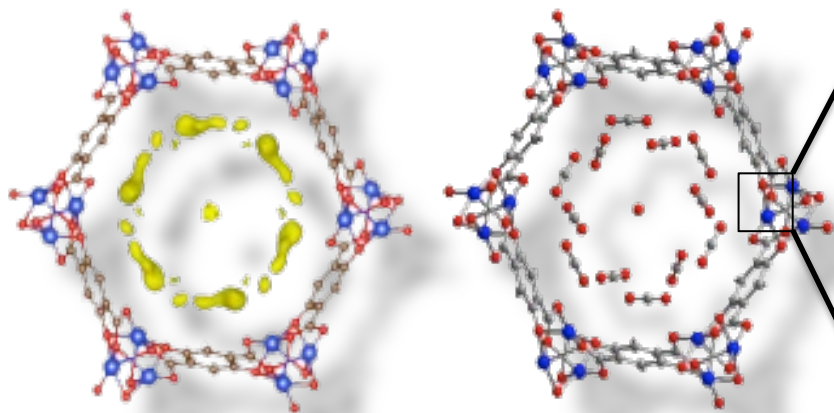
McKinsey & Co Reducing US GHG Emissions 2007; Sumida, Rogow, Mason, McDonald, Bloch, Herm, Bae, Long, *Chem. Rev.* **2012**, *112*, 724.; D'Alessandro, Smit, Long, *Angew. Chem. Int. Ed.*, **2010**, *49*, 6058; IEA Statistics CO₂ Emissions From Fuel Combustion, 2012.



Climate Action: Capture carbon dioxide in MOFs



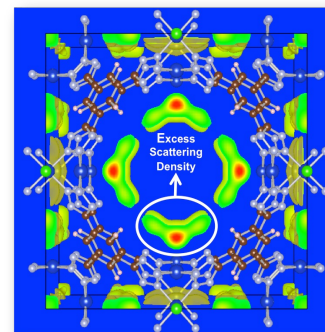
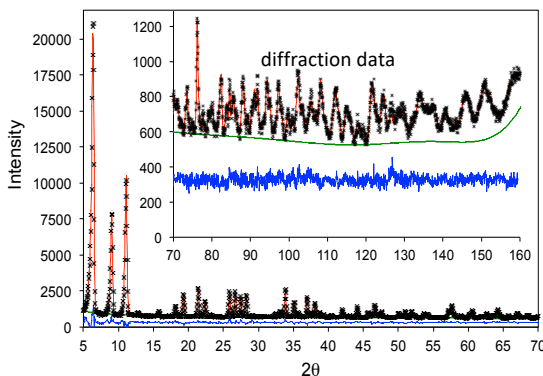
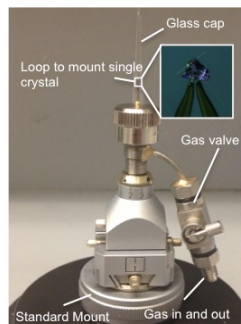
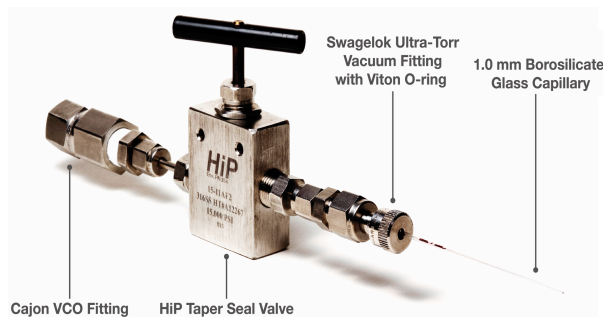
Mehrdad Asgari



In-situ characterization



Breakthrough apparatus



Chem. Sci., **2017**, 8, 43

Chem. Sci. **2014**, 5, 4569. Featured cover article.

Chem. Sci., **2018**, 9, 4579. Featured cover article.



Water contamination: One of the world's leading causes of death

11% of the world population has no access to clean drinking water

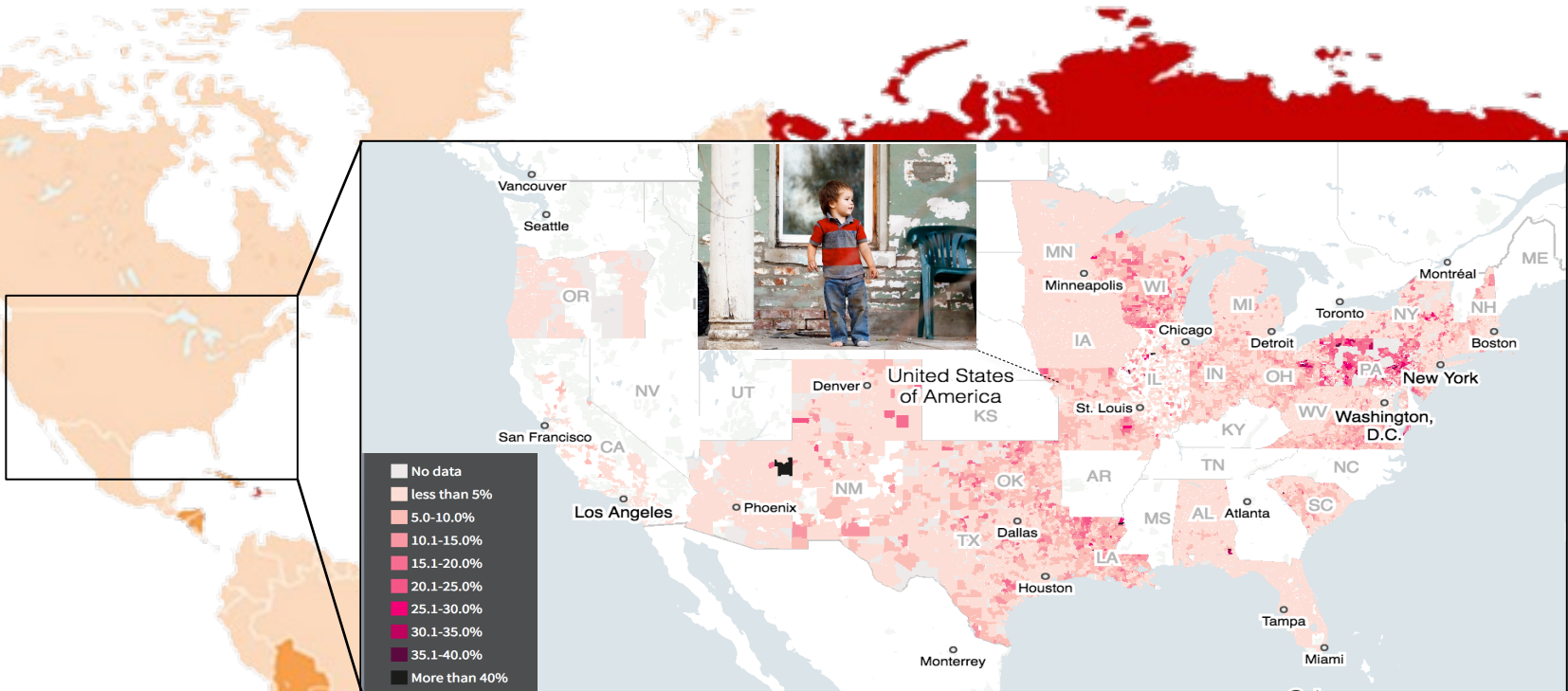


<http://www.scmp.com/news/china/article/1839337/beijing-drinking-water-reservoir-found-contain-levels-lead-20-times-who>
<http://www.cnn.com/2016/01/11/health/toxic-tap-water-flint-michigan/index.html>



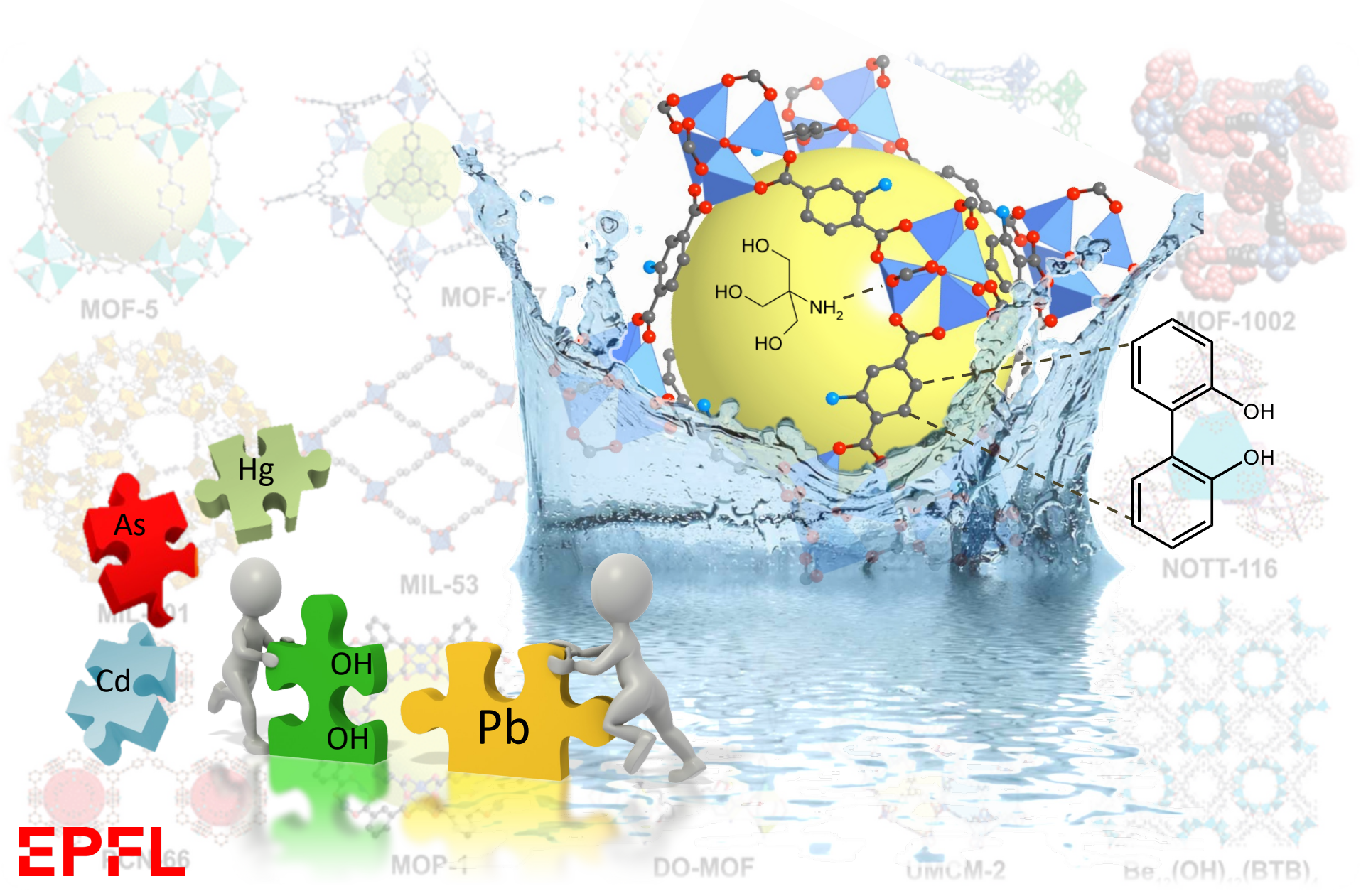
Water contamination: One of the world's leading causes of death

December 2016: Reuters shows 3000 areas with Pb levels twice that of Flint.



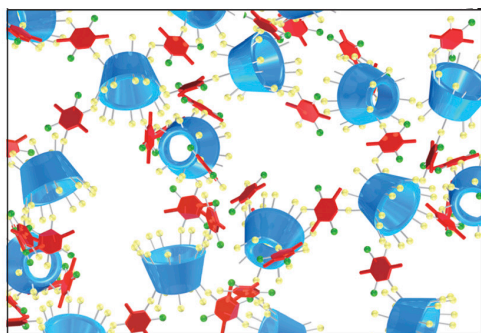
<http://www.reuters.com/investigates/special-report/usa-lead-testing/#interactive-lead>

Tuning MOFs for water purification

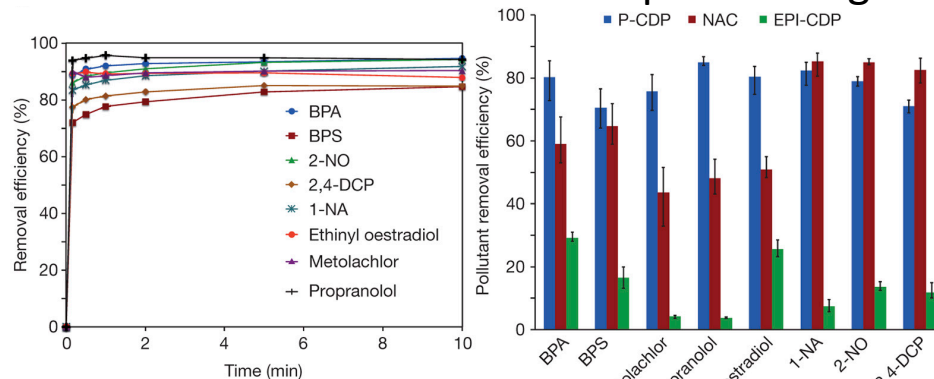


Functionalizing MOFs for enhanced performance

Porous polymers compete with commercial adsorbents for adsorption of organics

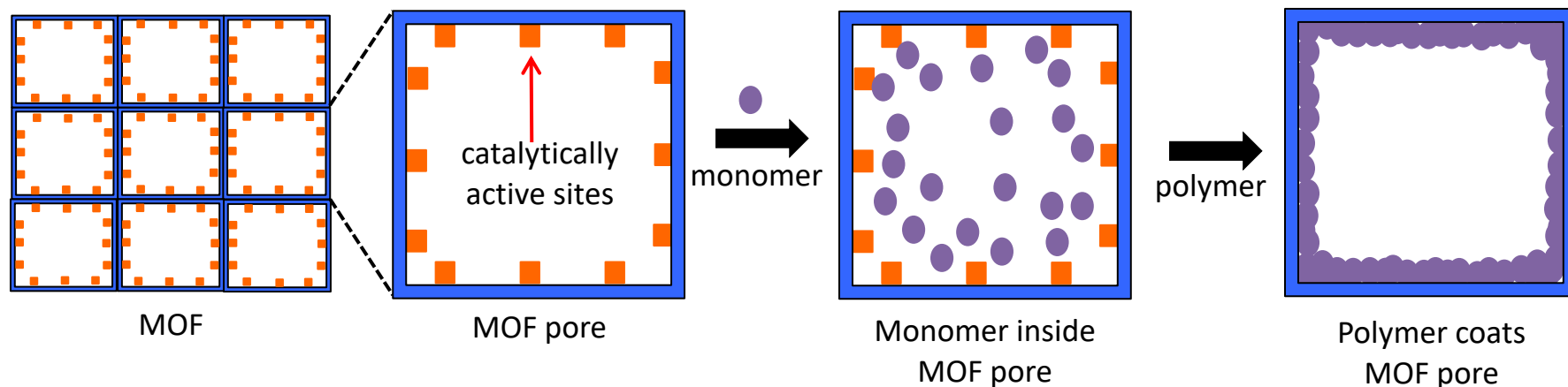


260 m²/g



Alsaiee, Smith, Xiao, Ling, Helbling, Dichtel, *Nature*, 529, 2016, 190–194.

Introducing functional porosity into intrinsically nonporous materials



■ - Open Metal Sites that activates monomer

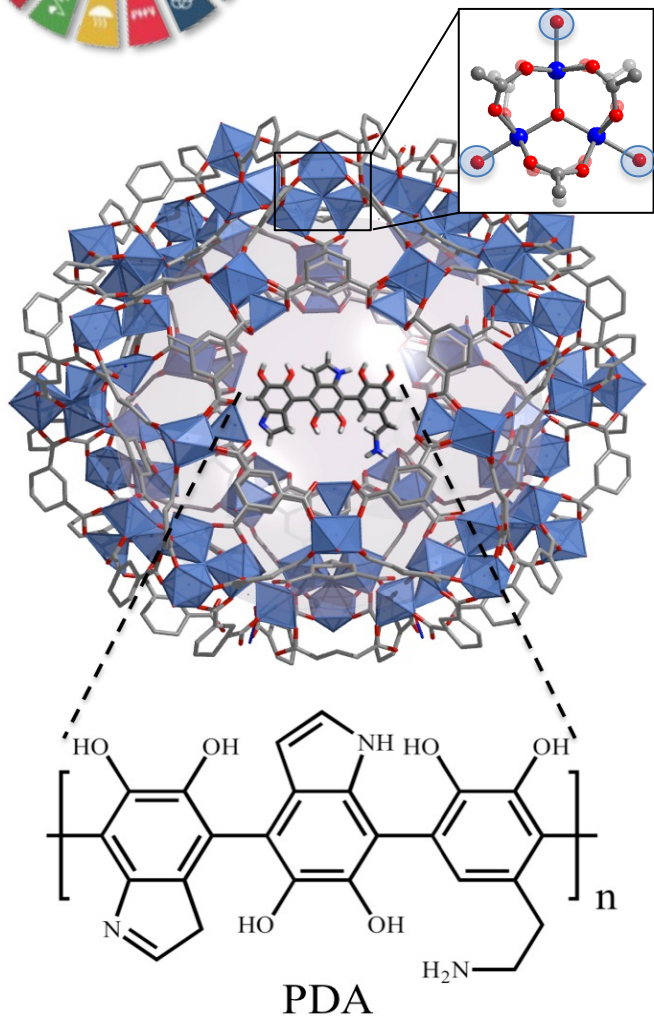
● - Polymer Building Block



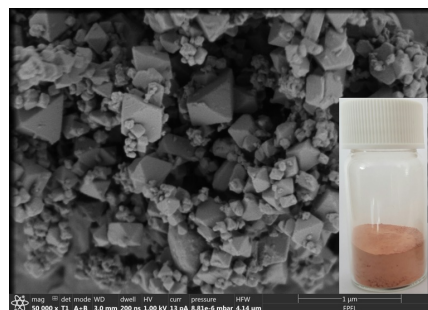
SDG 6

Clean water: Extracting heavy elements from water

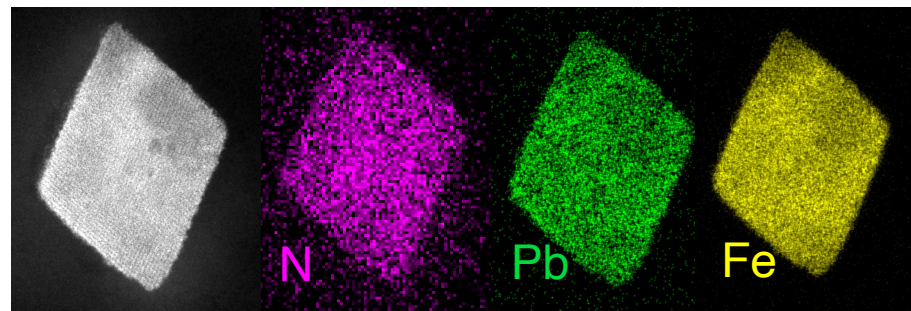
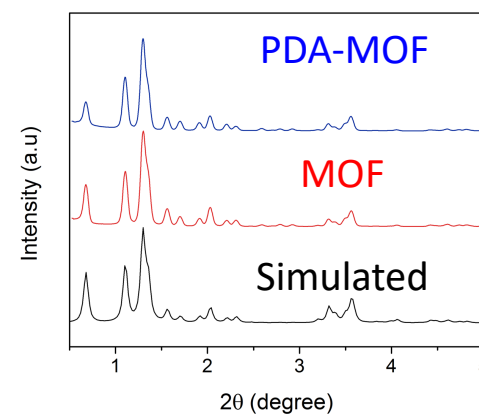
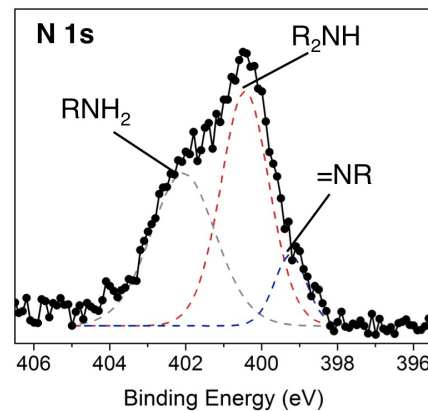
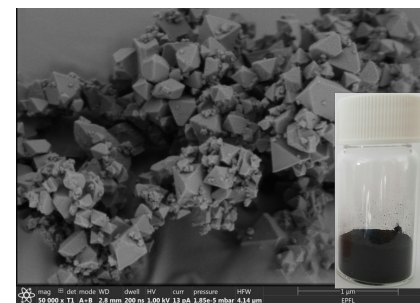
Introducing functional porosity into intrinsically nonporous materials



MOF



PDA-MOF



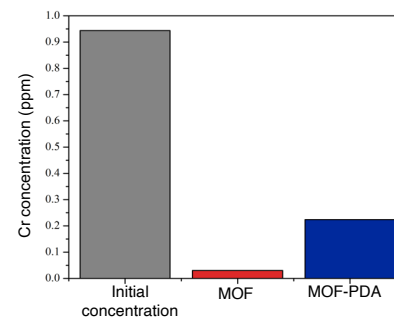
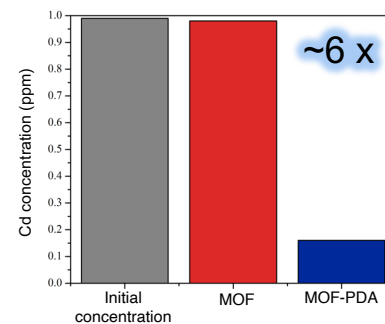
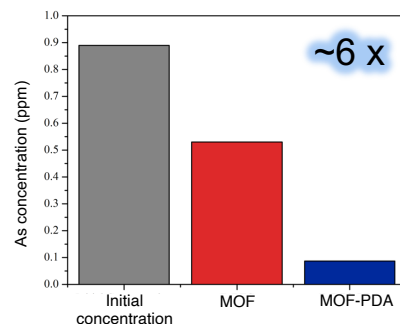
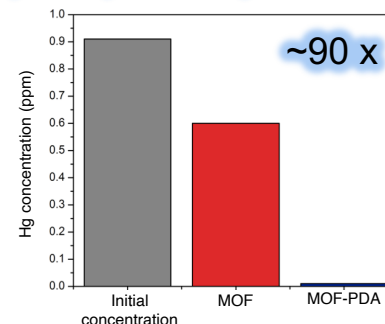
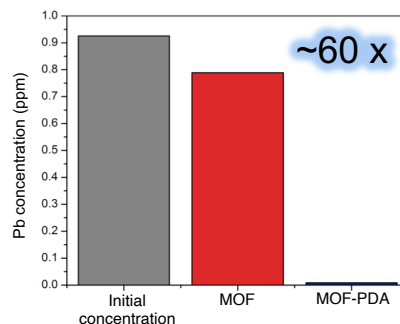
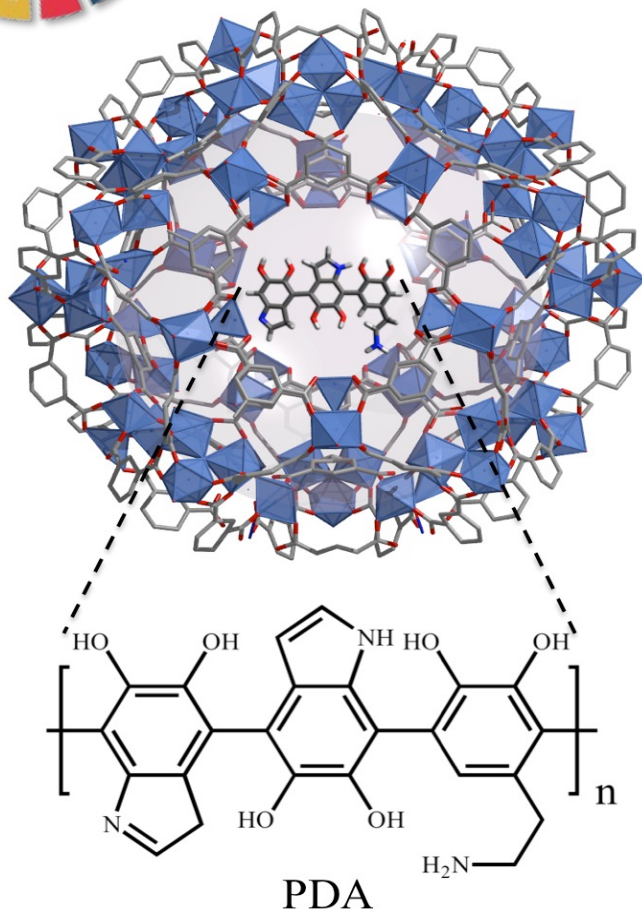
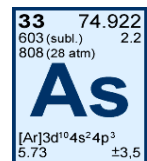
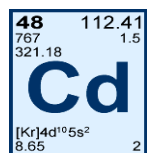
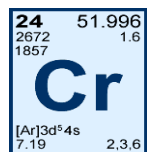
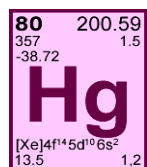
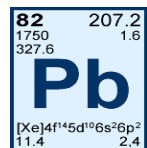


SDG 6

Clean water: Extracting heavy elements from water

Introducing functional porosity into intrinsically nonporous materials

Removal efficiency is improved by PDA



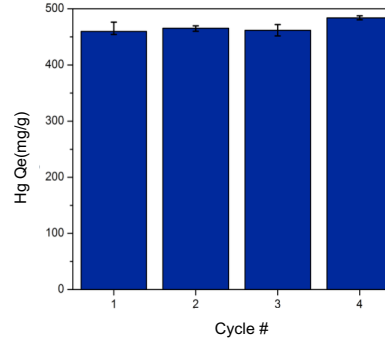
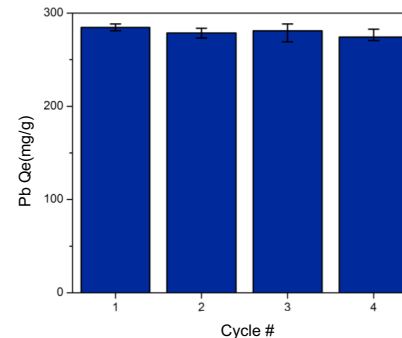
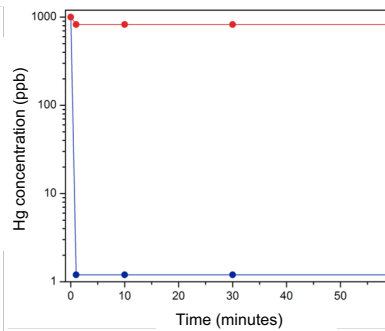
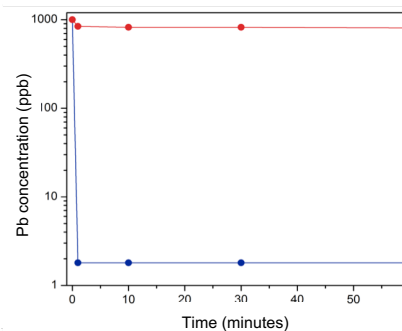
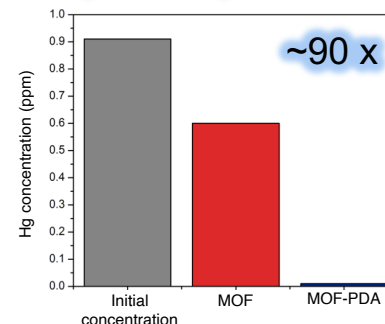
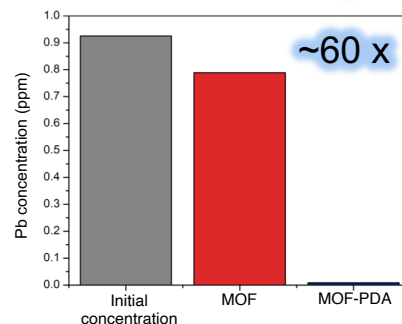
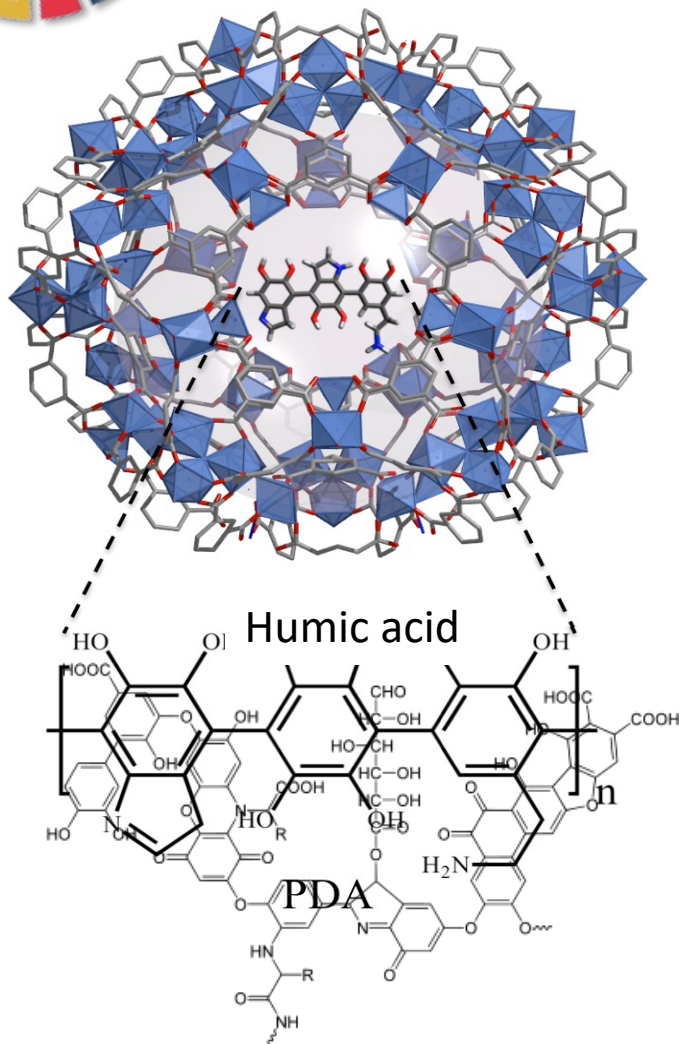
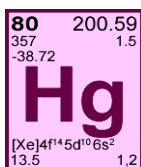
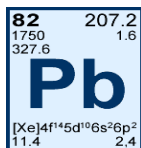


SDG 6

Clean water: Extracting mercury and lead from water

Introducing functional porosity into intrinsically nonporous materials

Removal capacity is improved by PDA

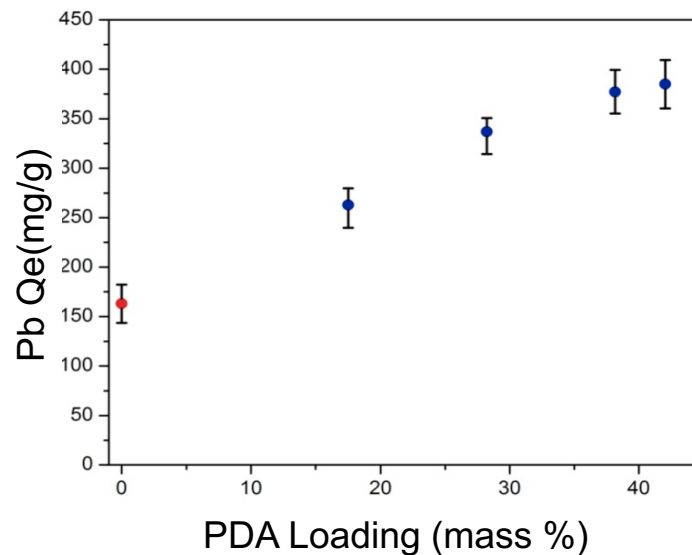
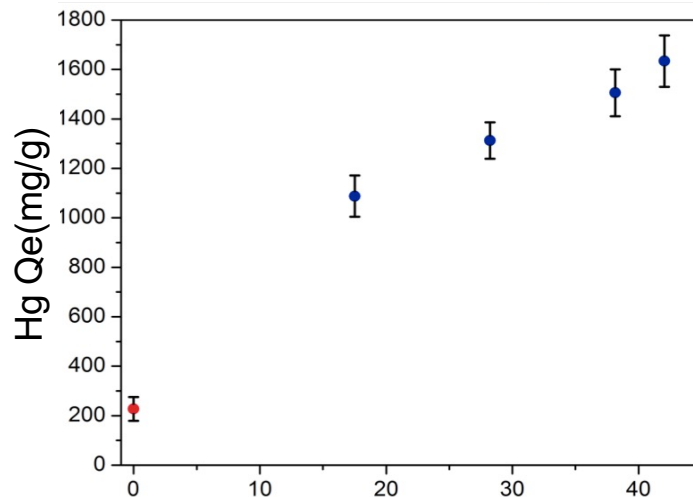
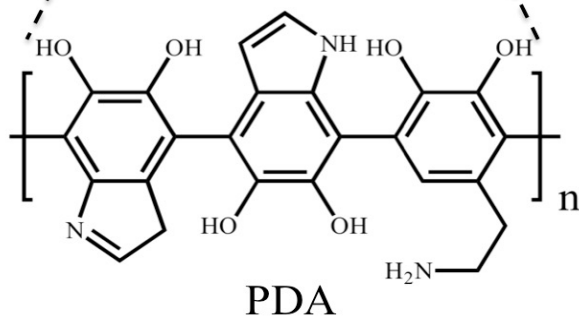
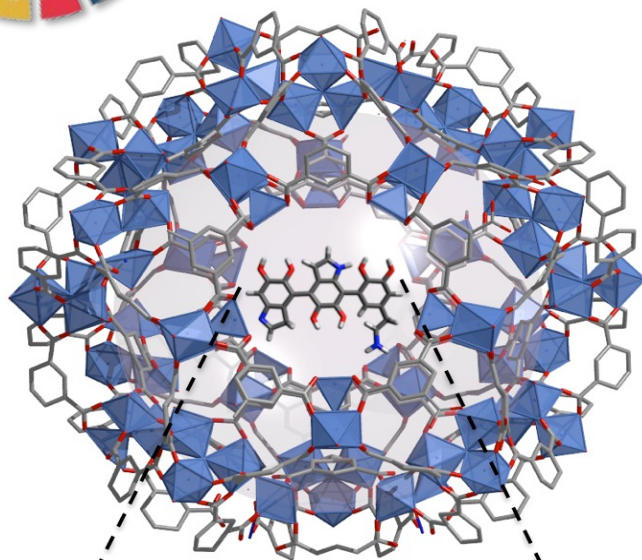
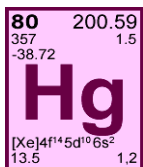
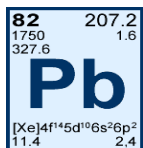




SDG 6

Clean water: Tuning dopamine leading

Introducing functional porosity into intrinsically nonporous materials



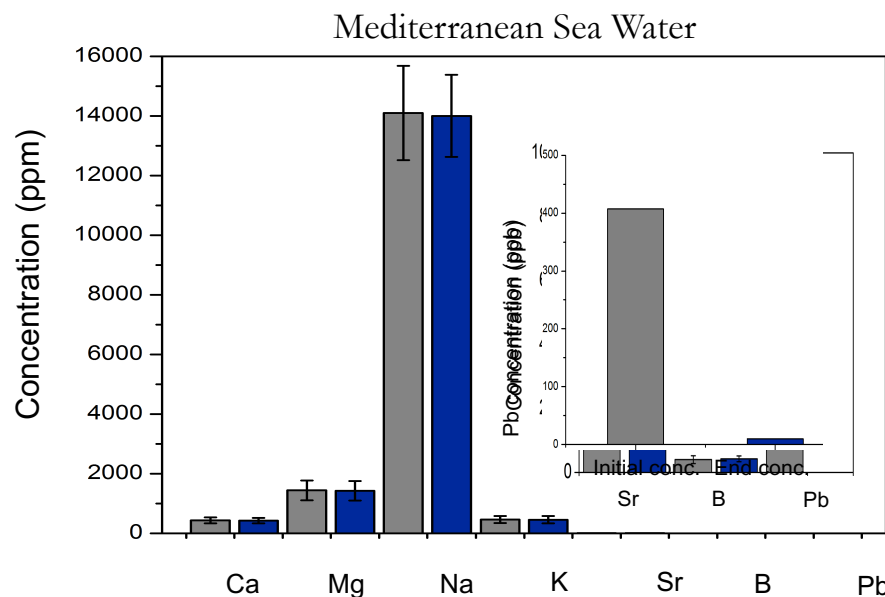
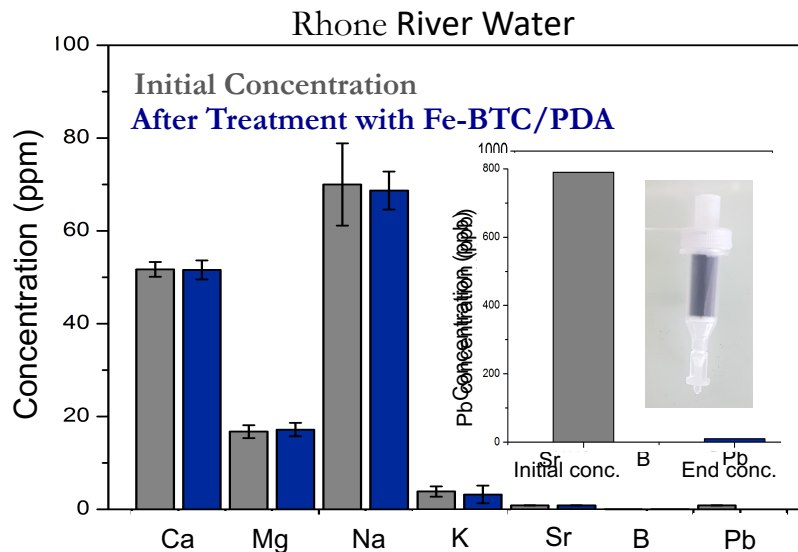
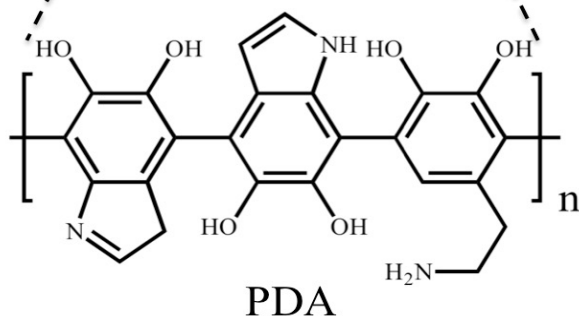
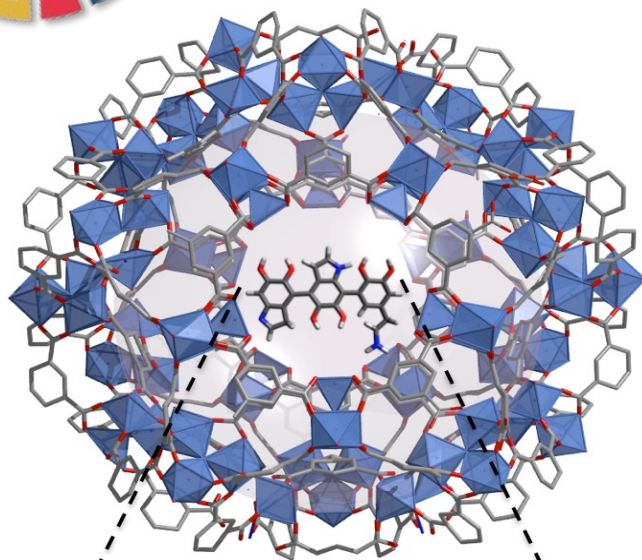
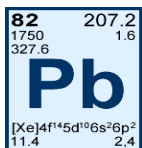
PDA Loading (mass %)



SDG 6

Clean water: Highly selective heavy metal extraction

Introducing functional porosity into intrinsically nonporous materials





Clean water: How do we compare?

- ✓ High capacities: 1690 and 400 mg/g for Hg^{2+} and Pb^{2+} , respectively.
- ✓ High k_d s: 5.5×10^6 for Hg^{2+} and 1.7×10^6 for Pb^{2+}

$$k_d = \frac{C_o - C_e}{C_e} \left(\frac{V}{m} \right)$$

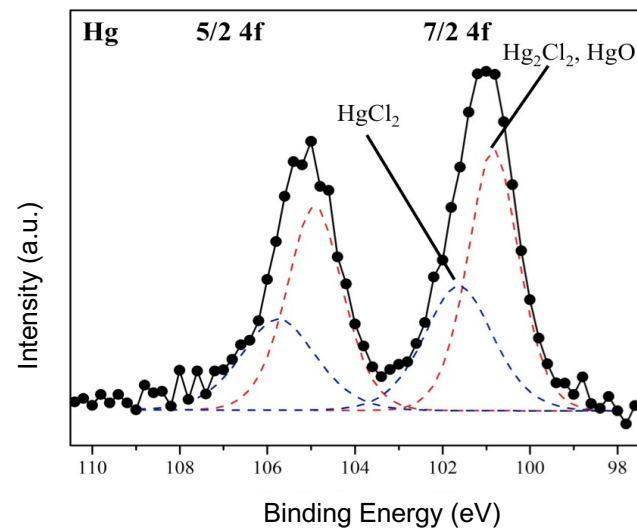
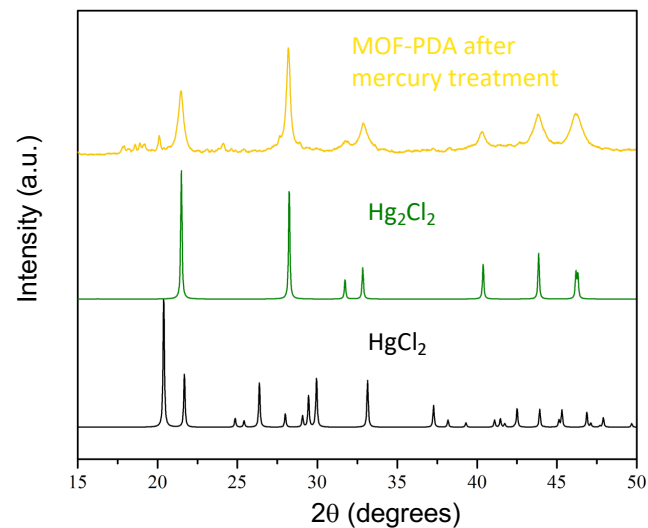
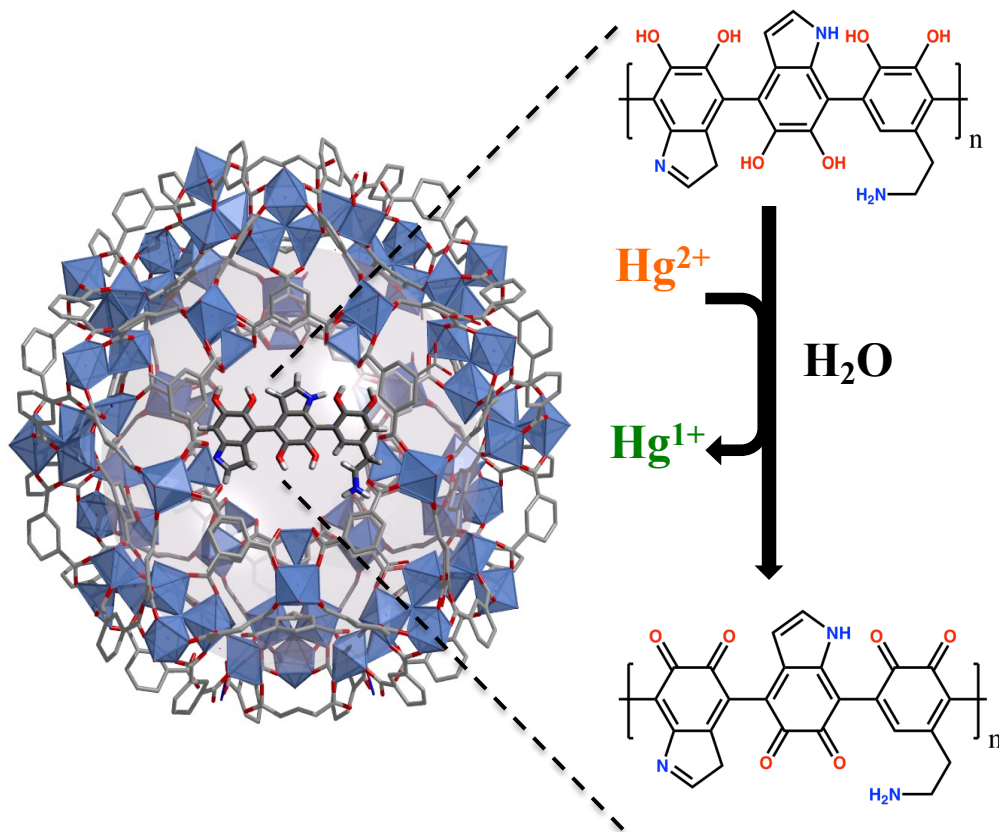
C_o = initial concentration, C_e = end concentration, V = volume (mL), m = mass (g)

- ✓ Cost: 2.50 USD per kg composite for starting materials
- ✓ Rates: a few seconds
- ✓ Reversibility: 5 cycles
- ✓ Stability: In water for several months
- ✓ Selectivity: resists clogging of common organic interferents and inorganics



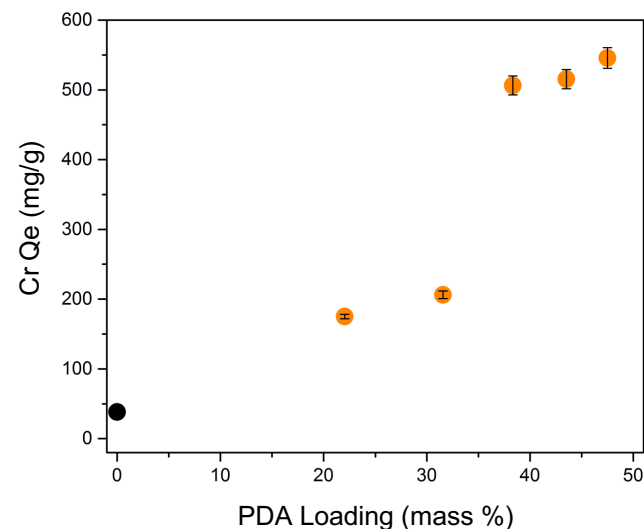
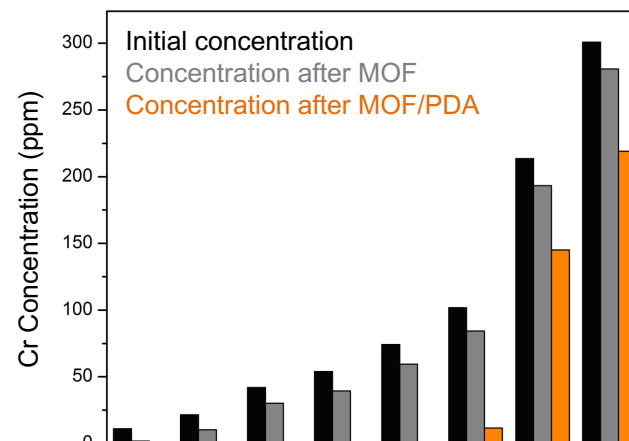
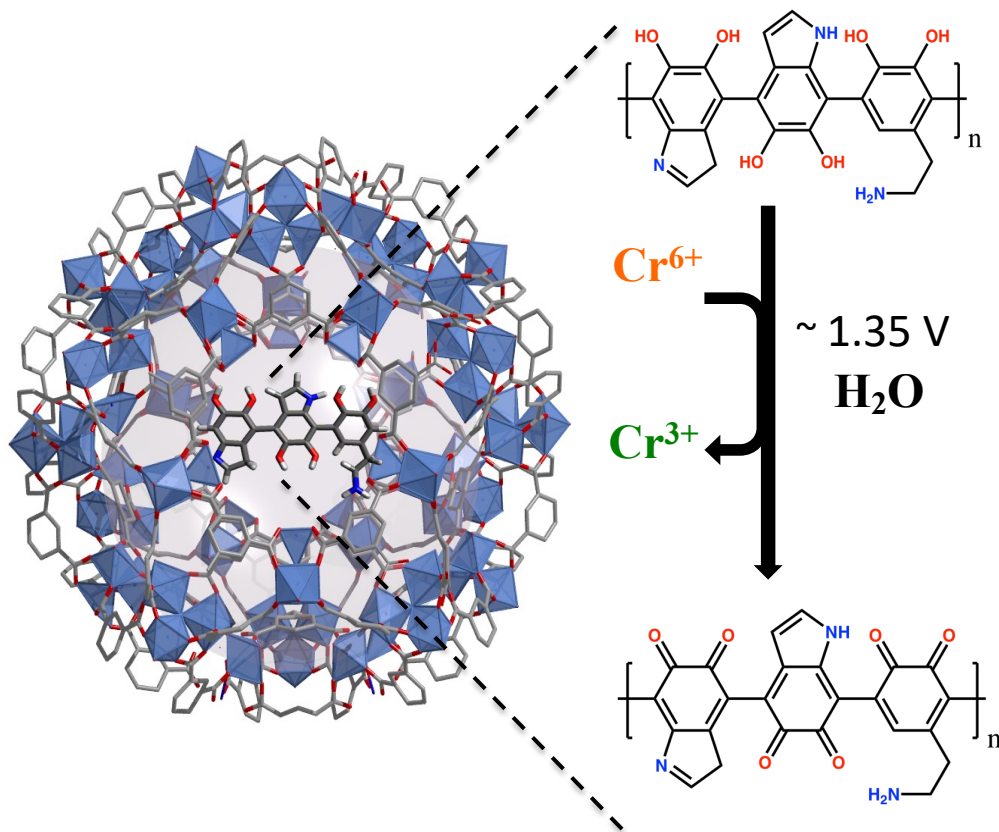
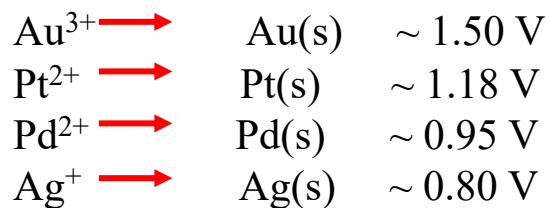


Clean water: Reduction of Hg^{2+}





Clean water: Reduction of Cr^{6+}



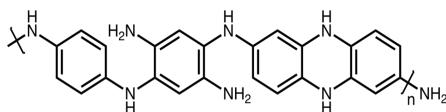


SDG 6

Other MOF/polymer composites

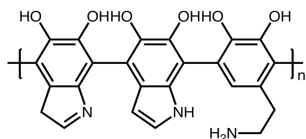
PpPDA

1800 m²/g



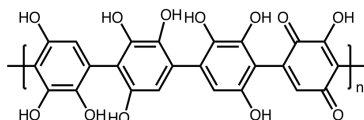
PDA

1100 m²/g



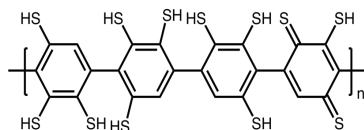
PHQ

900 m²/g



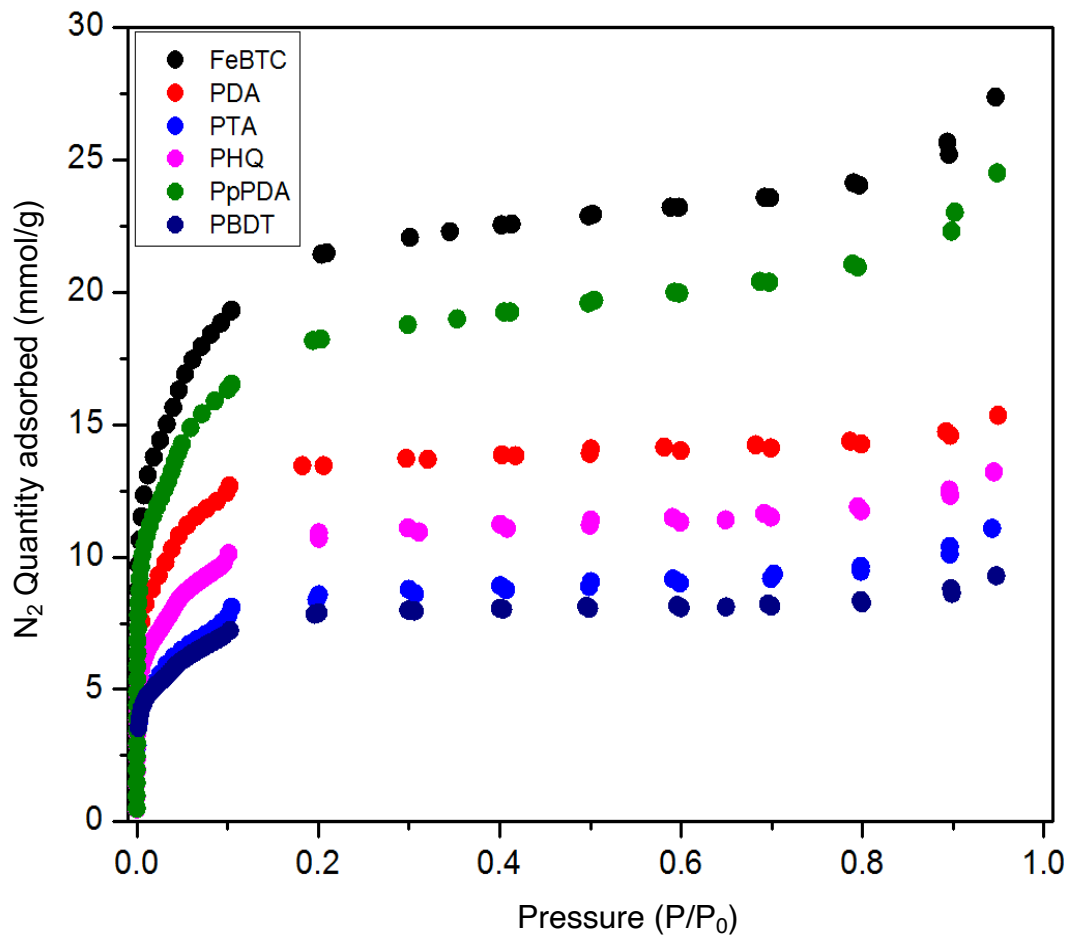
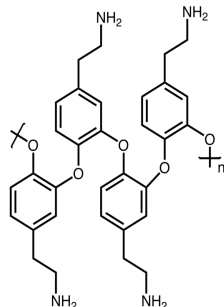
PBDT

650 m²/g



PTA

700 m²/g



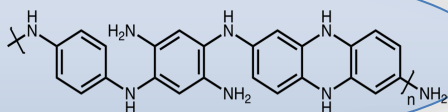


SDG 7

Responsible consumption: Gold extraction from river water

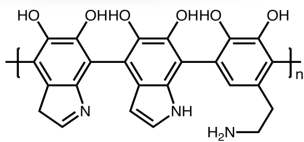
PpPDA

1800 m²/g



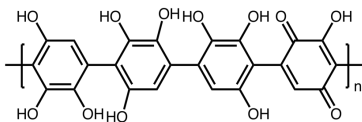
PDA

1100 m²/g



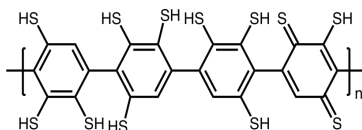
PHQ

900 m²/g



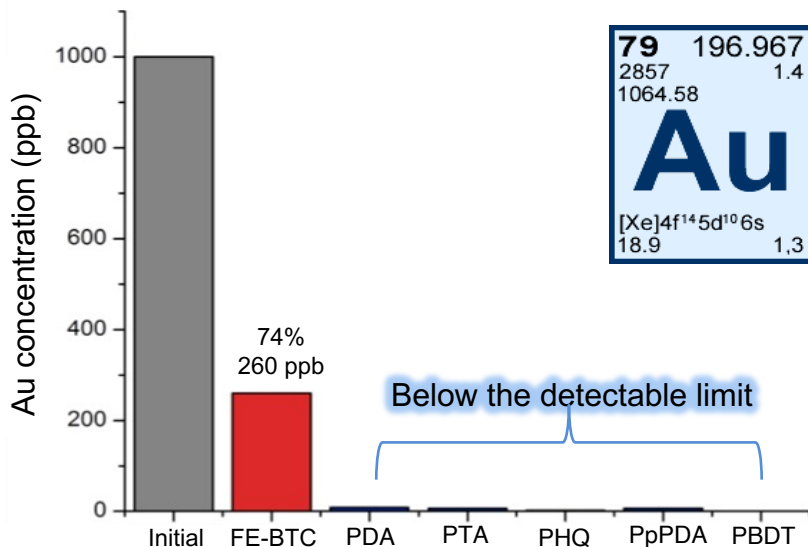
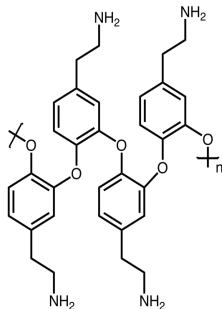
PBDT

650 m²/g

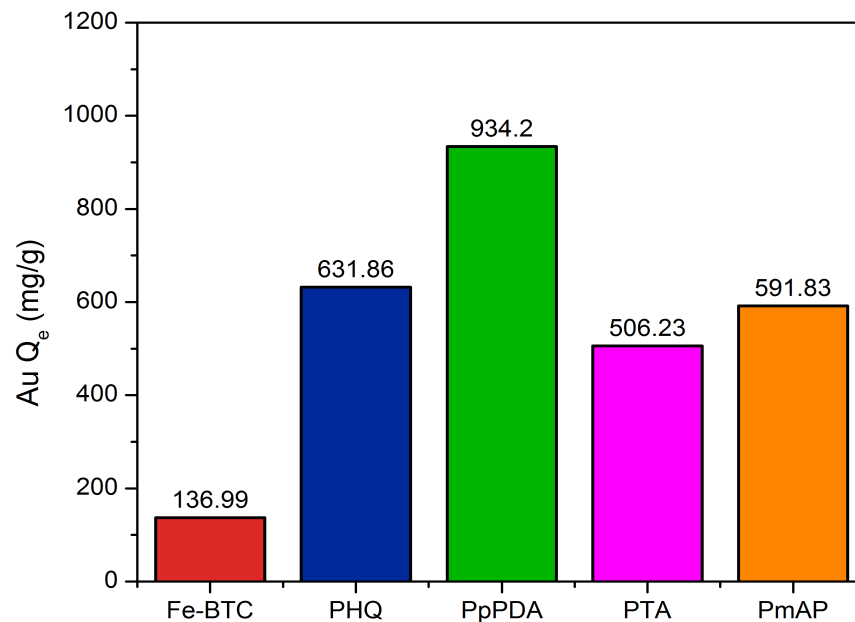


PTA

700 m²/g



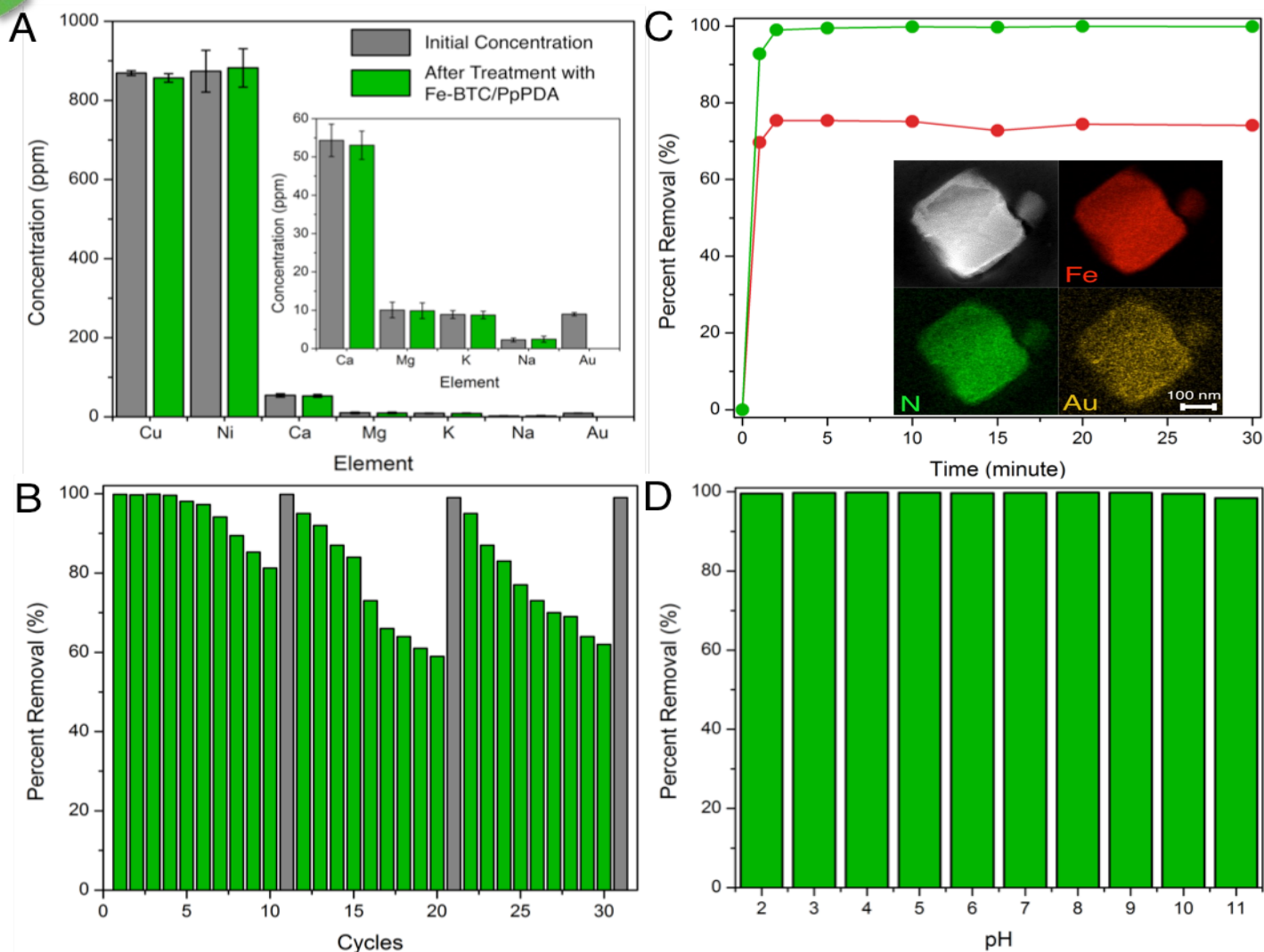
79	196.967
2857	1.4
1064.58	
Au	
[Xe]4f ¹⁴ 5d ¹⁰ 6s	
18.9	1.3





SDG 7

Responsible consumption: Gold extraction from river water

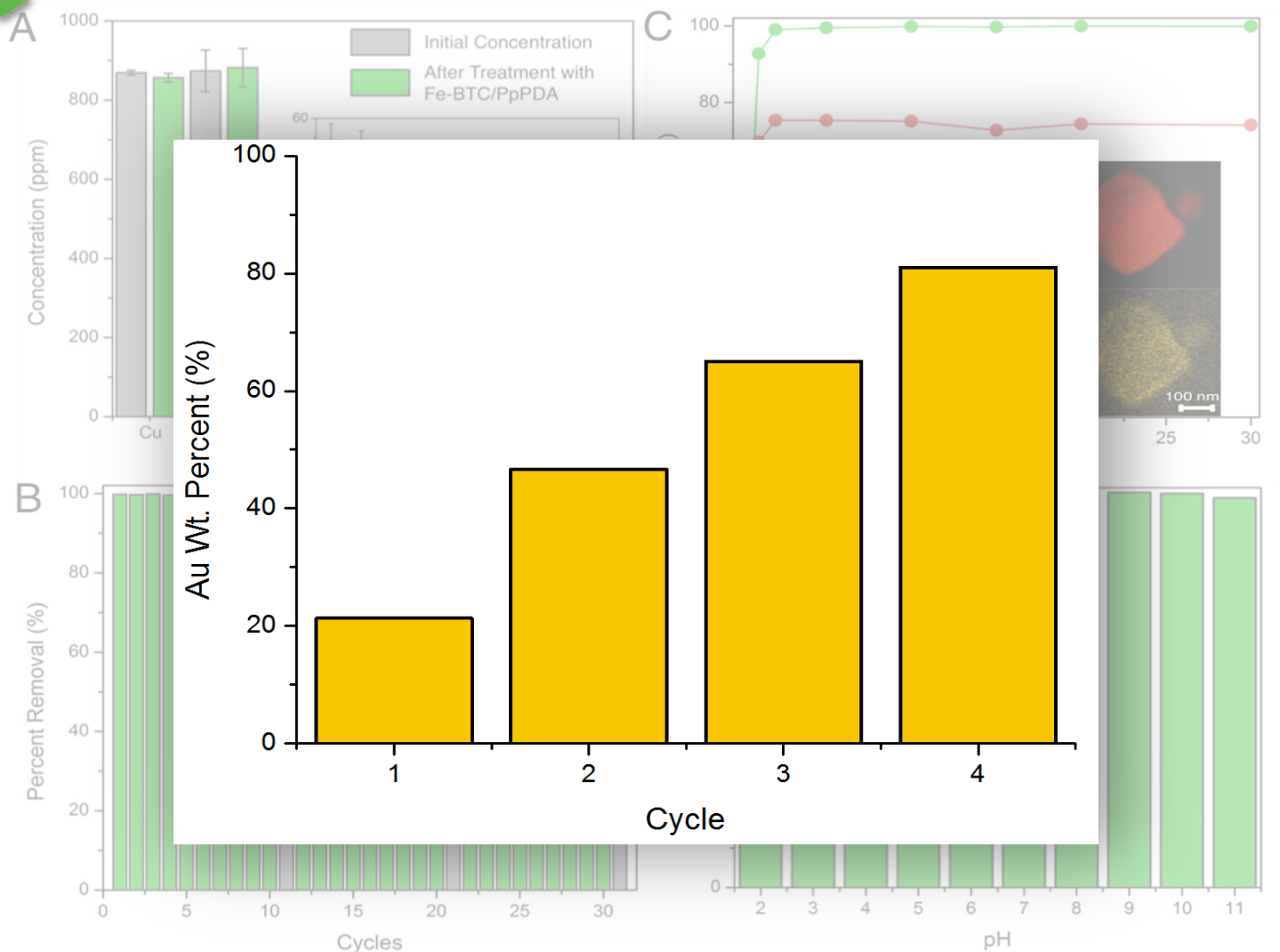


J. Am. Chem. Soc., **2018**, 140, 16697.
C&E News, 2018, 96, 47. ACS Editors Choice



SDG 7

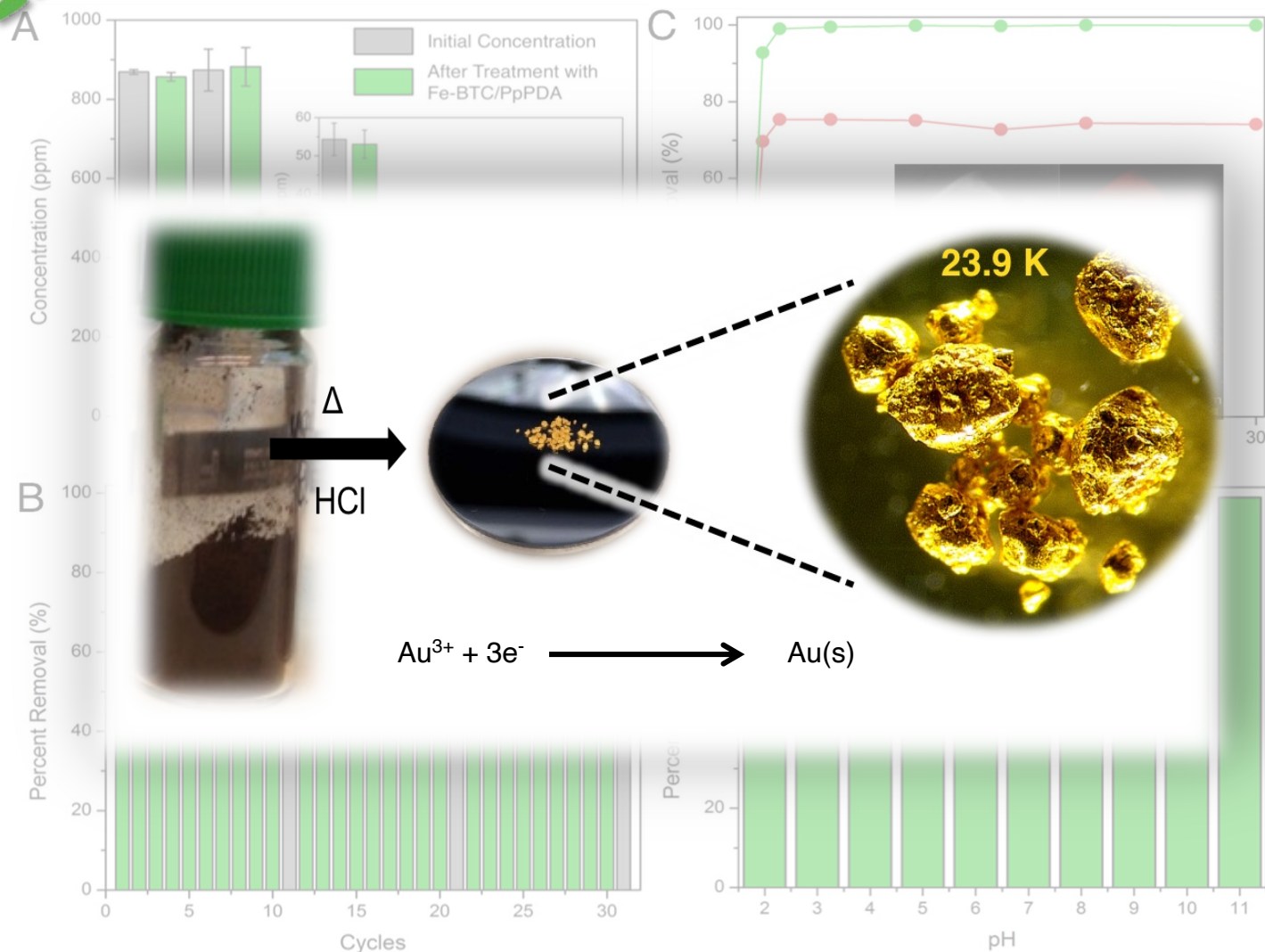
Responsible consumption: Gold extraction from river water





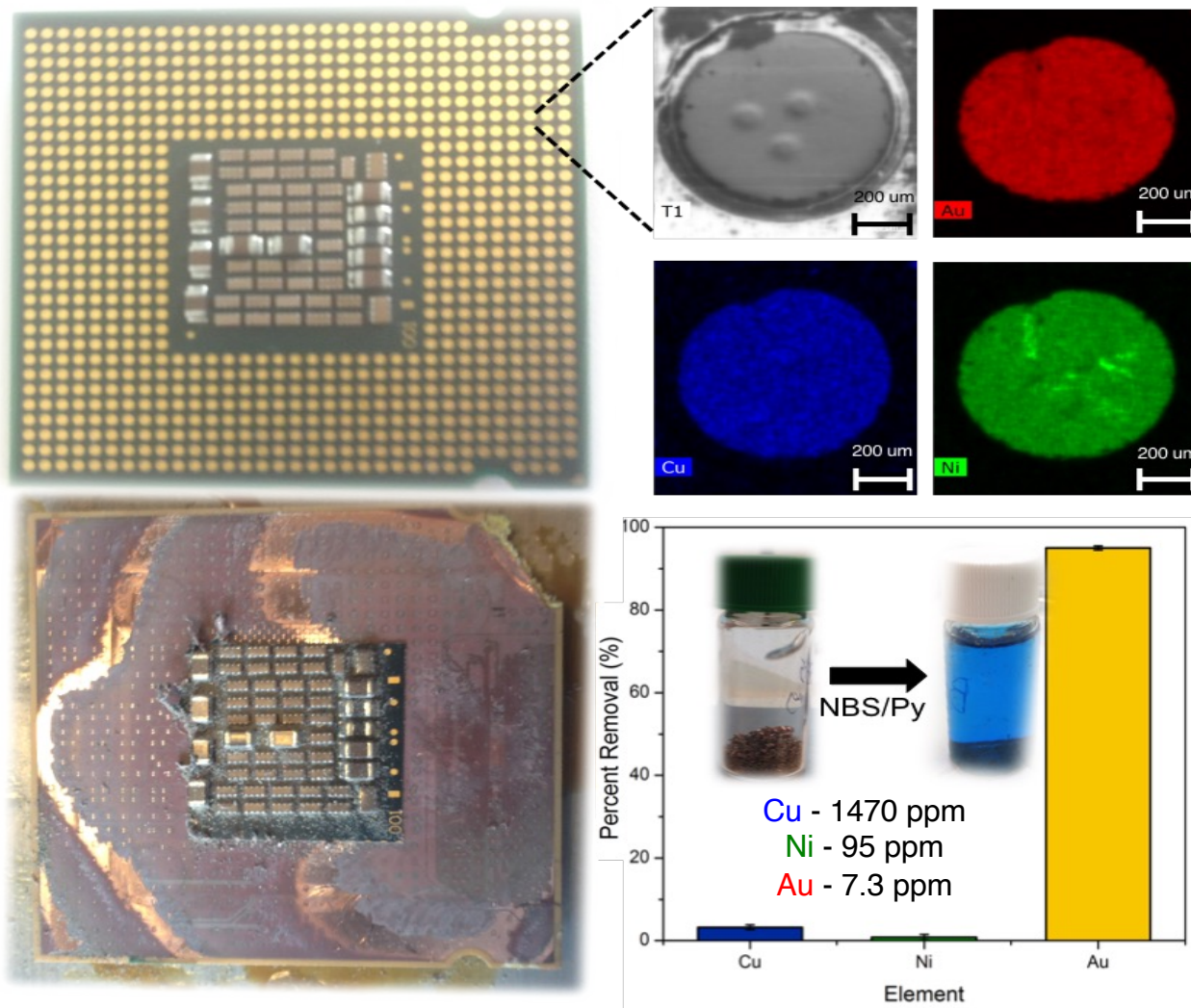
SDG 7

Responsible consumption: Gold extraction from river water





Responsible consumption: Gold recovery from electronic waste



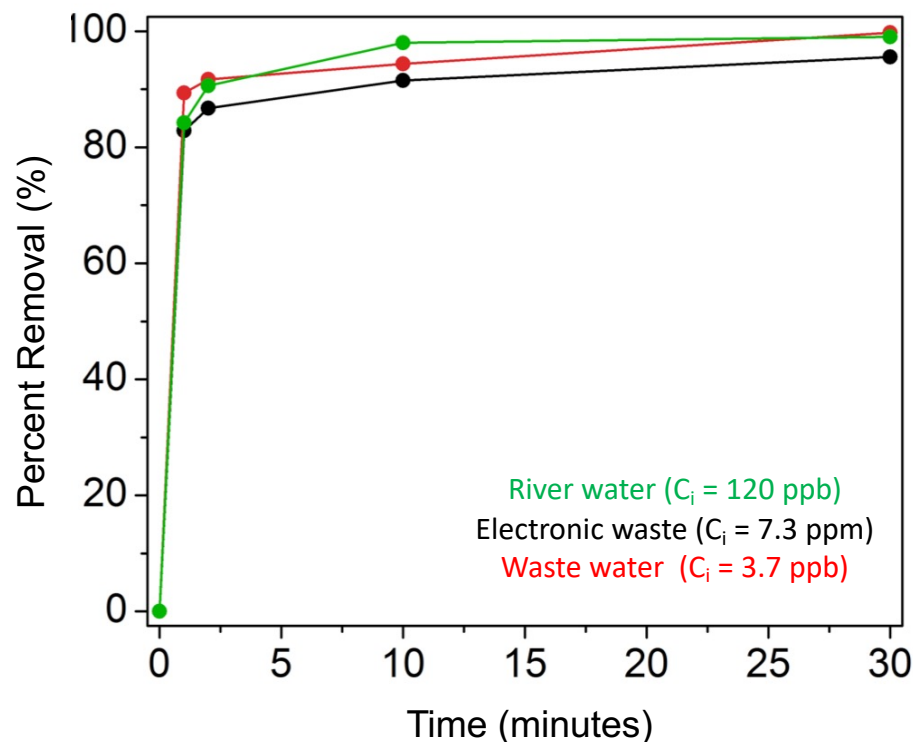
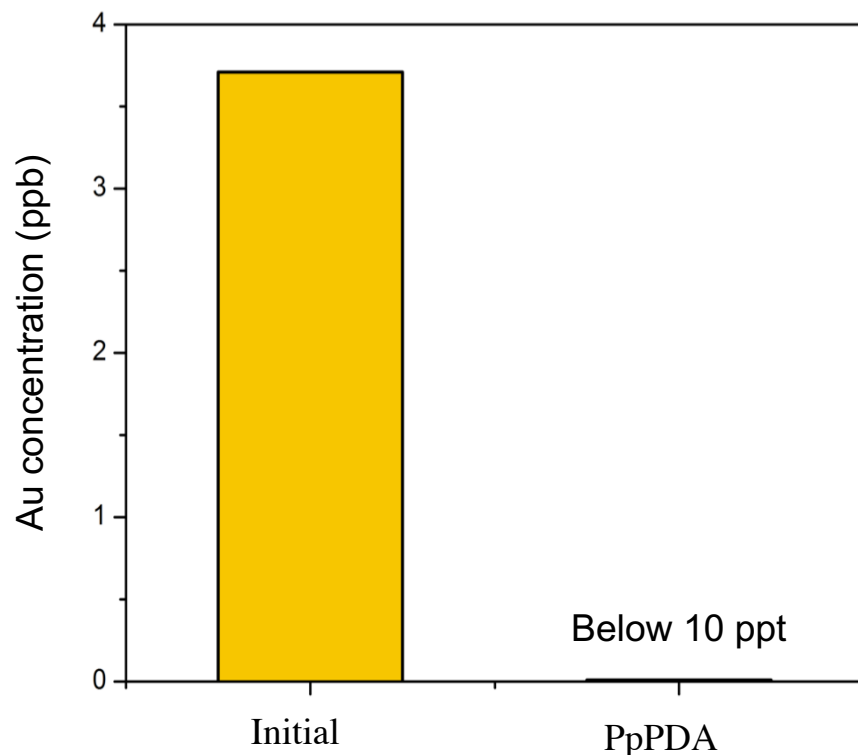


SDG 7

Responsible consumption: Gold recovery from wastewater

Scientists find gold worth \$2 million in Swiss sewage

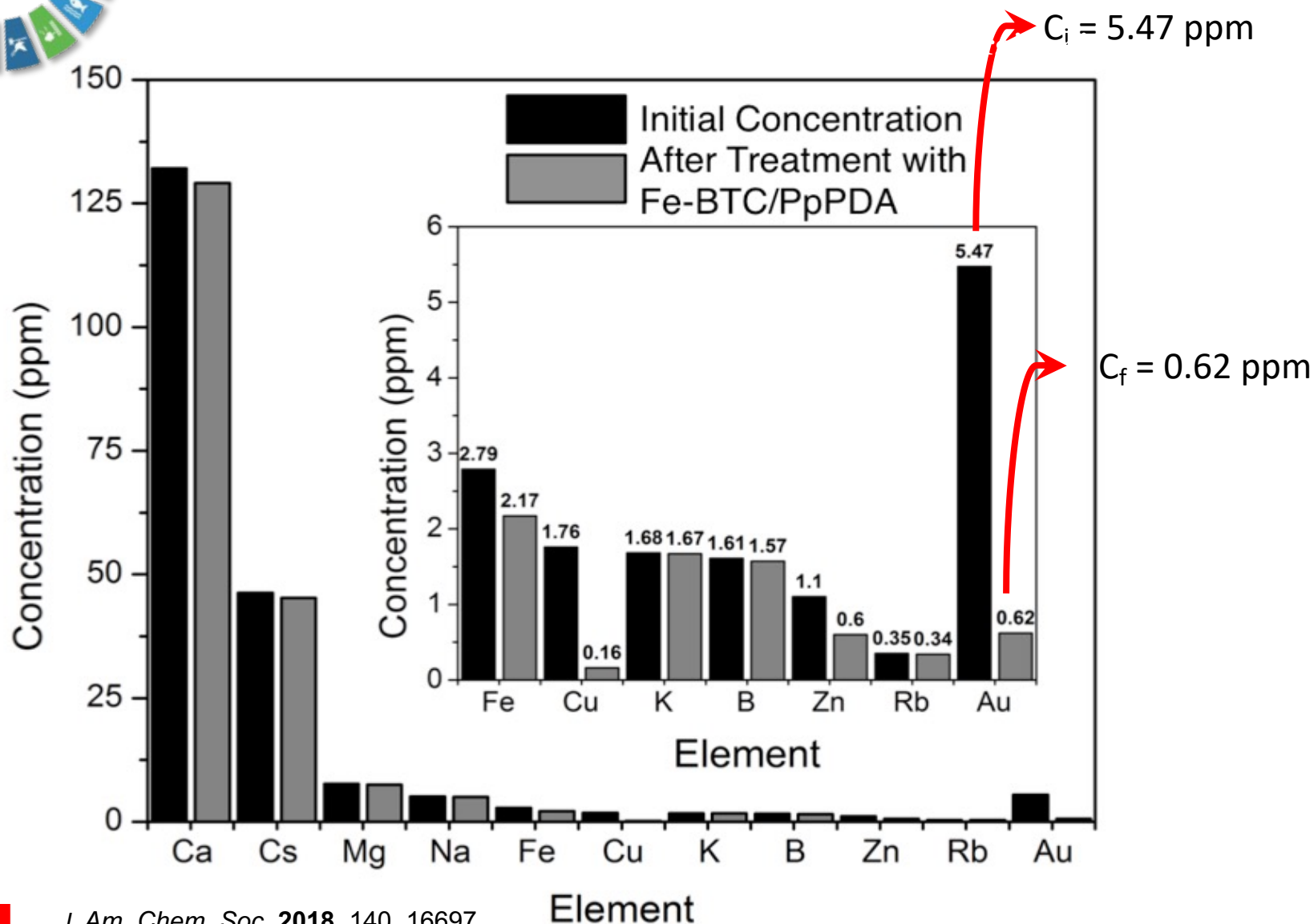
by Alanna Petroff @AlannaPetroff

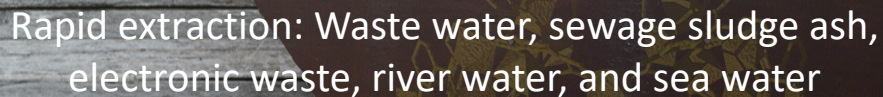
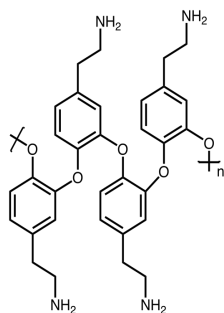
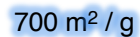
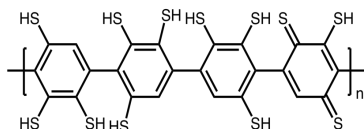
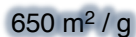
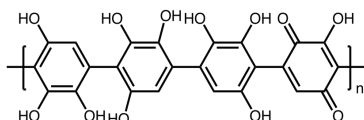
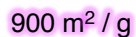
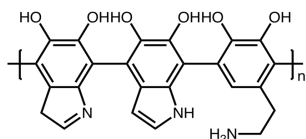
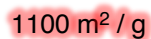
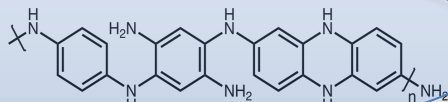
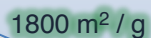




SDG 7

Responsible consumption: Gold recovery from sewage sludge ash





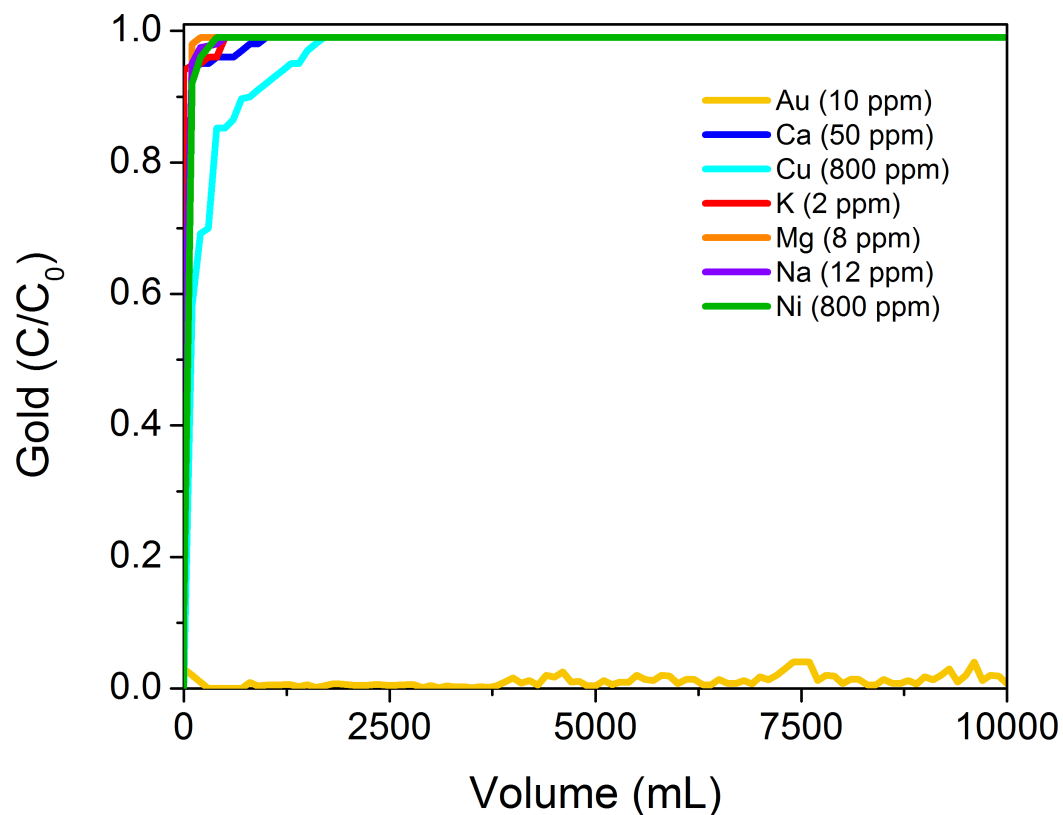
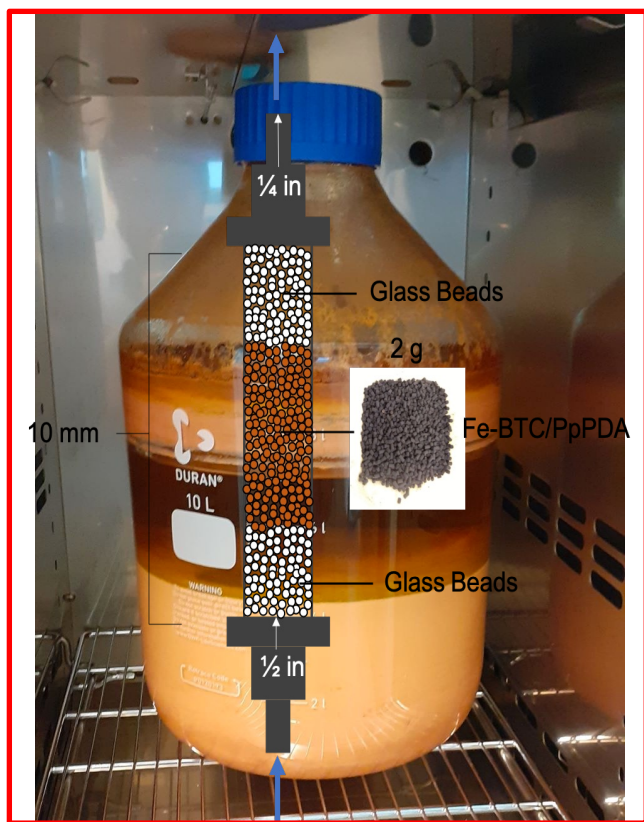


Where shall we go from here...

enable

600 gram synthesis in water,
reuse solvent

Installation of a flow through apparatus





Where shall we go from here...

1). Material design

Electronic Waste



Waste Water



Fresh Surface Water



Sea Water

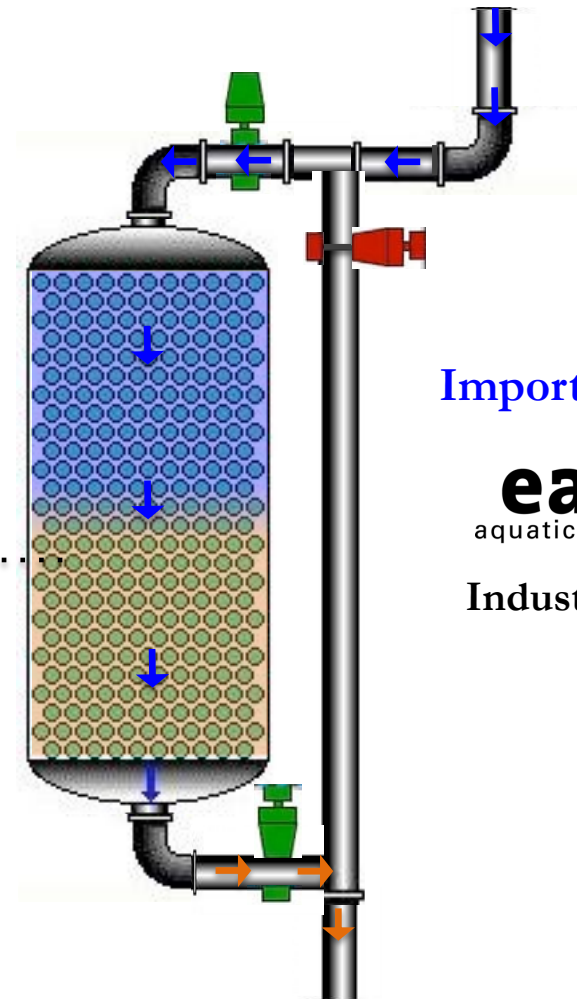


Sponge

2). Structuring



3). Scaled -up demonstration (kg) Flow contaminated mixture in



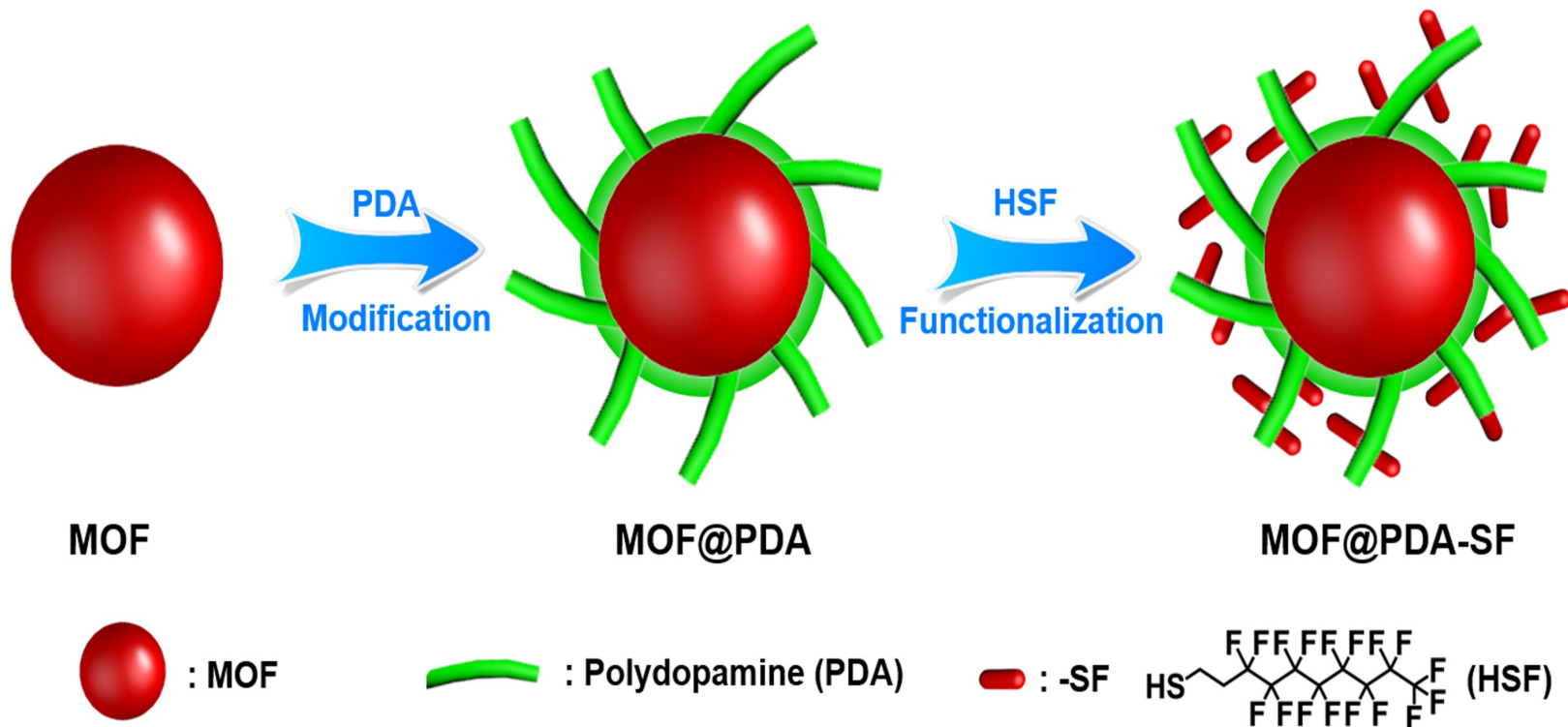
Important Partners

eawag
aquatic research ooo

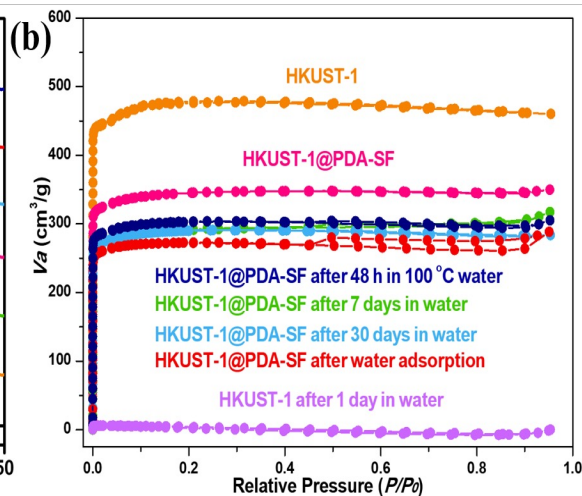
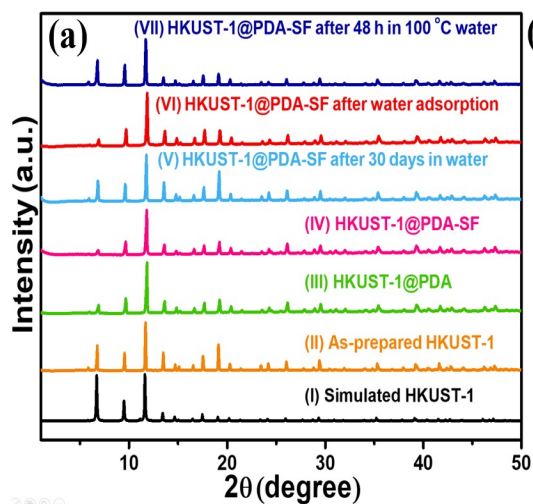
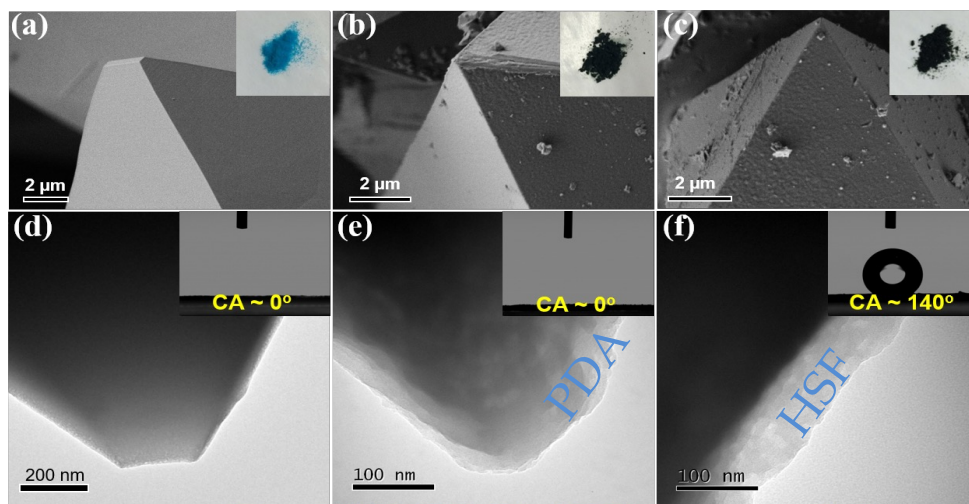
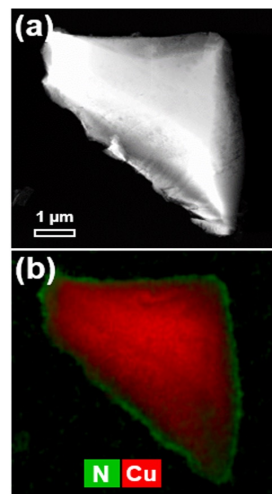
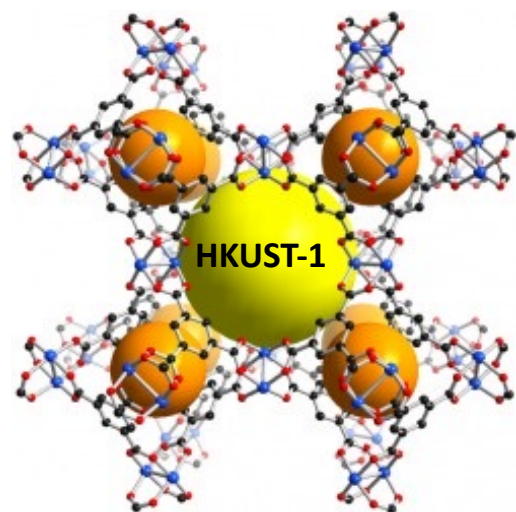
Industrial partners...

EPFL

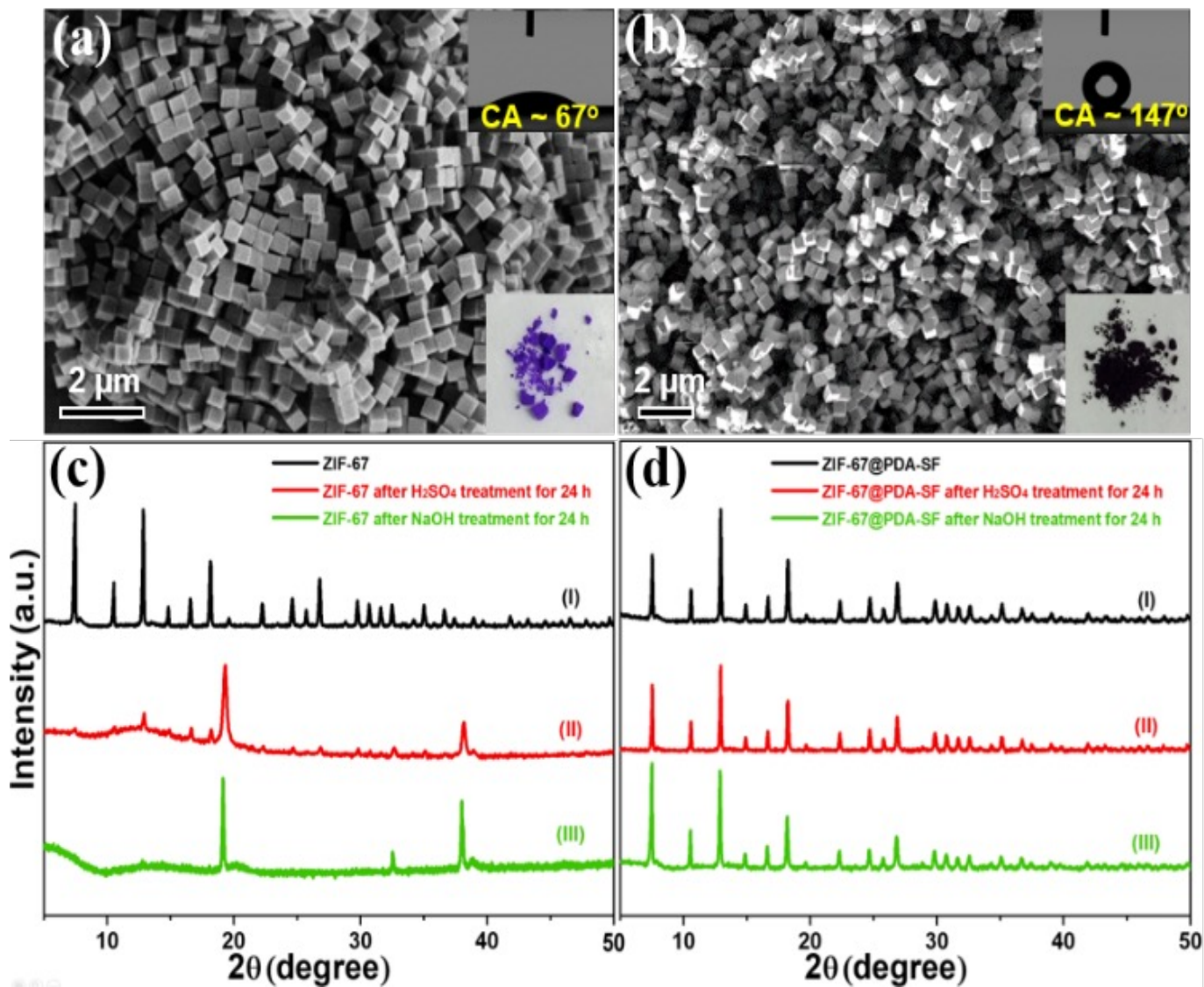
Chem Sci. **2019**, *10*, 4542. Featured cover article.



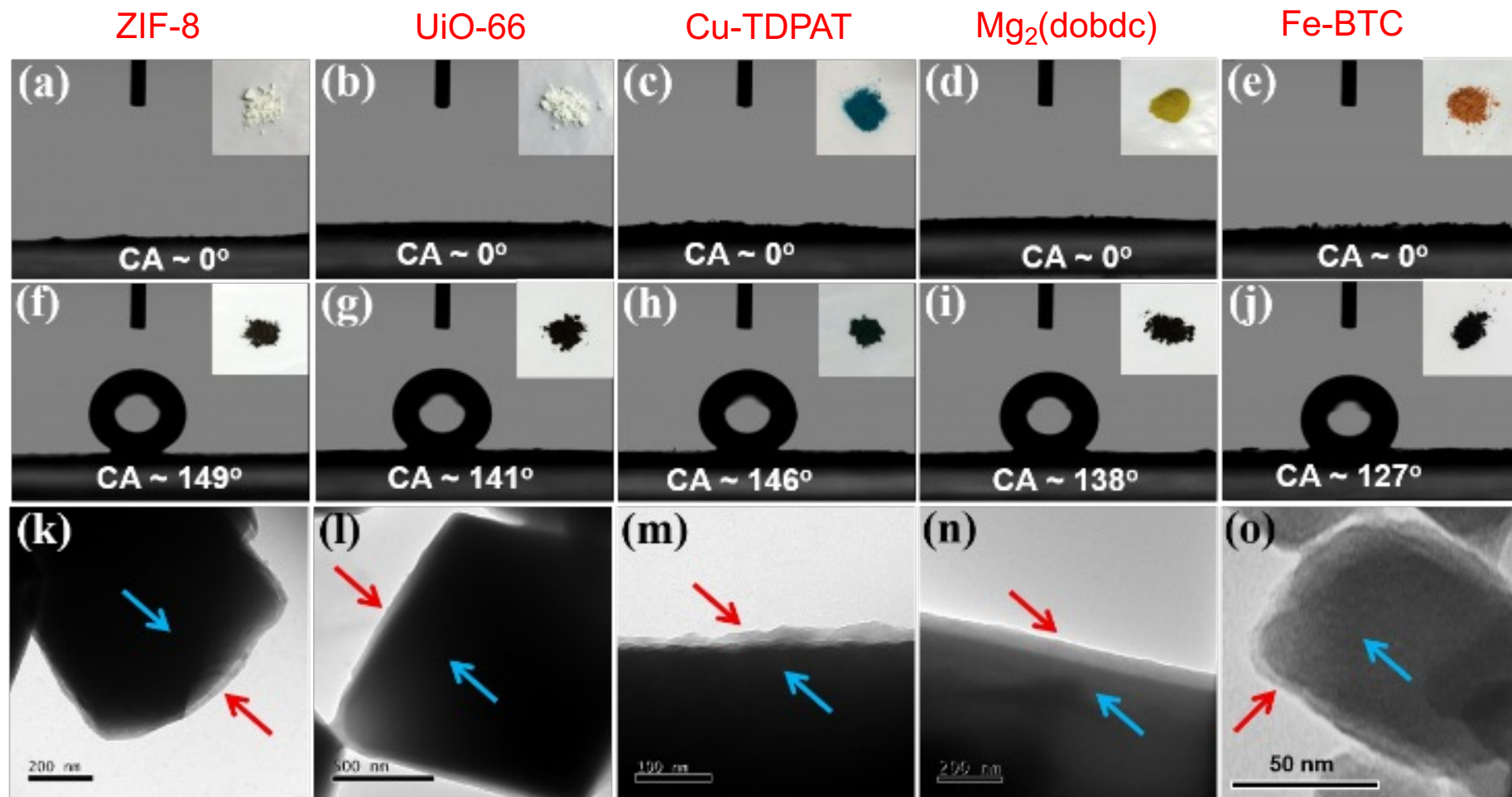
Hydrophobic coatings on HKUST-1



Hydrophobic coating on ZIF-67



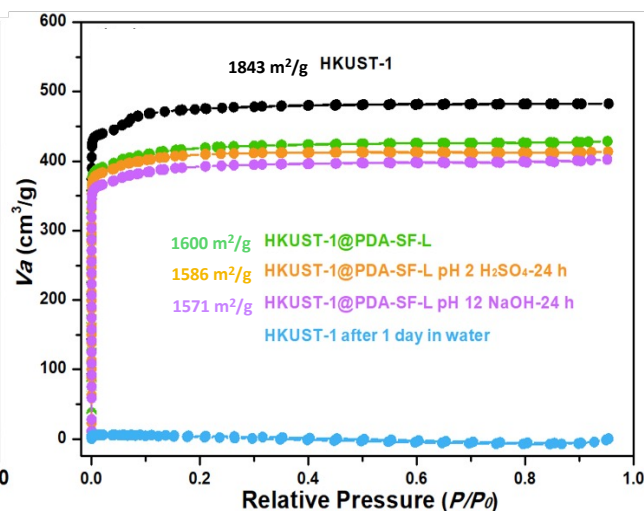
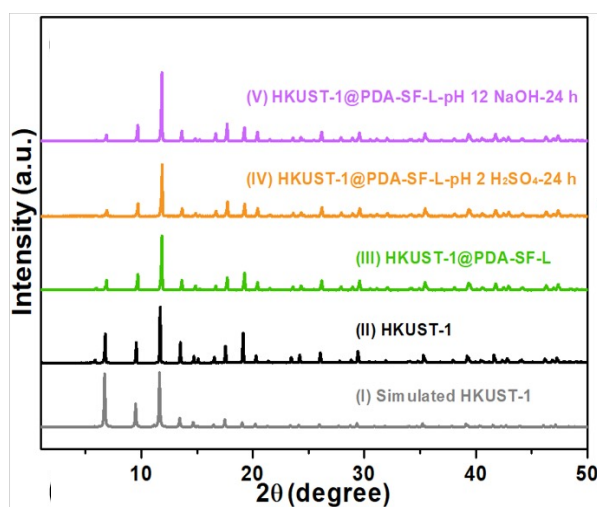
Tests on many structurally diverse MOFs



Optimizing surface coatings

Sample Name / Formula	BET Surface Area (m ² /g)	Contact Angle (°)	Polymer Loading (wt%)	H ₂ SO ₄ Treatment, pH = 0, 1, 2 or 3 for 24 h	BET Surface Area after H ₂ SO ₄ Treatment (m ² /g)	NaOH Treatment, pH = 11, 12 or 13 for 24 h	BET Surface Area after NaOH Treatment (m ² /g)
-----------------------	--------------------------------------	-------------------	-----------------------	--	---	--	---

6.8 wt % PDA



<i>UiO-66</i> Zr ₆ O ₄ (OH) ₄ (BDC) ₆ , BDC ²⁻ = 1,4-dicarboxybenzene	1430	~0	---	X	---	X	---
<i>UiO-66@PDA-SF</i>	638	141	10.3	✓	626	✓	581
<i>Cu-TDPAT</i> Cu ₃ (TDPAT), TDPAT=2,4,6-tris(3,5-dicarboxylphenylamino)-1,3,5-triazine	2254	~0	---	X	---	X	---
<i>Cu-TDPAT@PDA-SF</i>	1835	146	12.5	✓	1784	✓	1778

Tests on many structurally diverse MOFs

Sample Name / Formula	BET Surface Area (m ² /g)	Contact Angle (°)	Polymer Loading (wt%)	H ₂ SO ₄ Treatment, pH = 0, 1, 2 or 3 for 24 h	BET Surface Area after H ₂ SO ₄ Treatment (m ² /g)	NaOH Treatment, pH = 11, 12 or 13 for 24 h	BET Surface Area after NaOH Treatment (m ² /g)
HKUST-1 Cu ₃ (BTC) ₂ , BTC ³⁻ = benzene-1,3,5-tricarboxylate	1854	~0	---	X	---	X	---
HKUST-1@PDA-SF	1286	140	16.8	✓	1247	✓	1242
ZIF-67 Co(MIm) ₂ , MIm=2-methylimidazolate	1466	67	---	X	---	X	---
ZIF-67@PDA-SF	683	147	16.8	✓	729	✓	738
ZIF-8 Zn(MIm) ₂ , MIm=2-methylimidazolate	1788	~0	---	X	---	✓	1710
ZIF-8@PDA-SF	1389	149	7.6	✓	1387	✓	1411
Mg-MOF-74 Mg ₂ (dobdc), dobdc ⁴⁻ = 2,5-dioxido-1,4-benzenedicarboxylate	1182	~0	---	X	---	✓	1138
Mg-MOF-74@PDA-SF	918	138	12.8	✓	867	✓	952
MIL-100-Fe Fe ₃ (BTC) ₂ , BTC ³⁻ = benzene-1,3,5-tricarboxylate	1587	~0	---	X	---	✓	1579
MIL-100-Fe@PDA-SF	1266	127	14.7	✓	1369	✓	1262
UiO-66 Zr ₆ O ₄ (OH) ₄ (BDC) ₆ , BDC ²⁻ = 1,4-dicarboxybenzene	1430	~0	---	X	---	X	---
UiO-66@PDA-SF	638	141	10.3	✓	626	✓	581
Cu-TDPAT Cu ₃ (TDPAT), TDPAT=2,4,6-tris(3,5-dicarboxylphenylamino)-1,3,5-triazine	2254	~0	---	X	---	X	---
Cu-TDPAT@PDA-SF	1835	146	12.5	✓	1784	✓	1778

Optimizing surface coatings

Sample Name / Formula	BET Surface Area (m ² /g)	Contact Angle (°)	Polymer Loading (wt%)	H ₂ SO ₄ Treatment, pH = 0, 1, 2 or 3 for 24 h	BET Surface Area after H ₂ SO ₄ Treatment (m ² /g)	NaOH Treatment, pH = 11, 12 or 13 for 24 h	BET Surface Area after NaOH Treatment (m ² /g)
HKUST-1 Cu ₃ (BTC) ₂ , BTC ³⁻ = benzene-1,3,5-tricarboxylate	1843	~0	---	X	---	X	---
HKUST-1@PDA-SF	1286	140	16.2	✓	1247	✓	1242
HKUST-1@PDA-SF-2	1600	135	6.8	✓	1586	✓	1571
ZIF-67 Co(MIm) ₂ , MIm=2-methylimidazolate	1466	67	---	X	---	X	---
ZIF-67@PDA-SF	683	147	16.8	✓	729	✓	738
ZIF-67@PDA-SF-2	1312	139	11.5	✓	1303	✓	1293
ZIF-8 Zn(MIm) ₂ , MIm=2-methylimidazolate	1788	~0	---	X	---	✓	1710
ZIF-8@PDA-SF	1389	149	7.6	✓	1387	✓	1411
Mg-MOF-74 Mg ₂ (dobdc), dobdc ⁴⁻ = 2,5-dioxido-1,4-benzenedicarboxylate	1182	~0	---	X	---	✓	1138
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UiO-66 Zr ₆ O ₄ (OH) ₄ (BDC) ₆ , BDC ²⁻ = 1,4-dicarboxybenzene	1430	~0	---	X	---	X	---
UiO-66@PDA-SF	638	144	12.2	✓	622	✓	594

Optimizing surface coatings

Chemical Science

Volume 10 | Number 17 | 7 May 2019 | Pages 4533–4716

rsc.li/chemical-science



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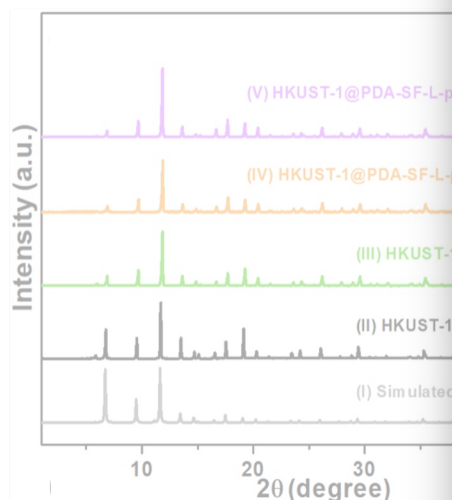
ROYAL SOCIETY OF CHEMISTRY | Celebrating IYPT 2019

EDGE ARTICLE
Wendy L. Queen et al.
A new post-synthetic polymerization strategy makes metal-organic frameworks more stable

Chem. Sci., 2019, 10, 4542. Featured cover article.

Sample Name

HKUST-1
Cu₃(BTC)₂, BTC³⁻
1,3,5-tricarboxybenzoate



Mg₂(dobdc), do
dioxido-
benzenedicarboxylate

Mg-MOF-74@

MIL-100
Fe₃(BTC)₂, BTC³⁻
1,3,5-tricarboxybenzoate

MIL-100-Fe@

UIO-66
Zr₆O₄(OH)₄(BDC)₆
dicarboxybenzoate

UIO-66@PDA-SF

BET
Surface
Area after
NaOH
Treatment
(m²/g)

1138

952

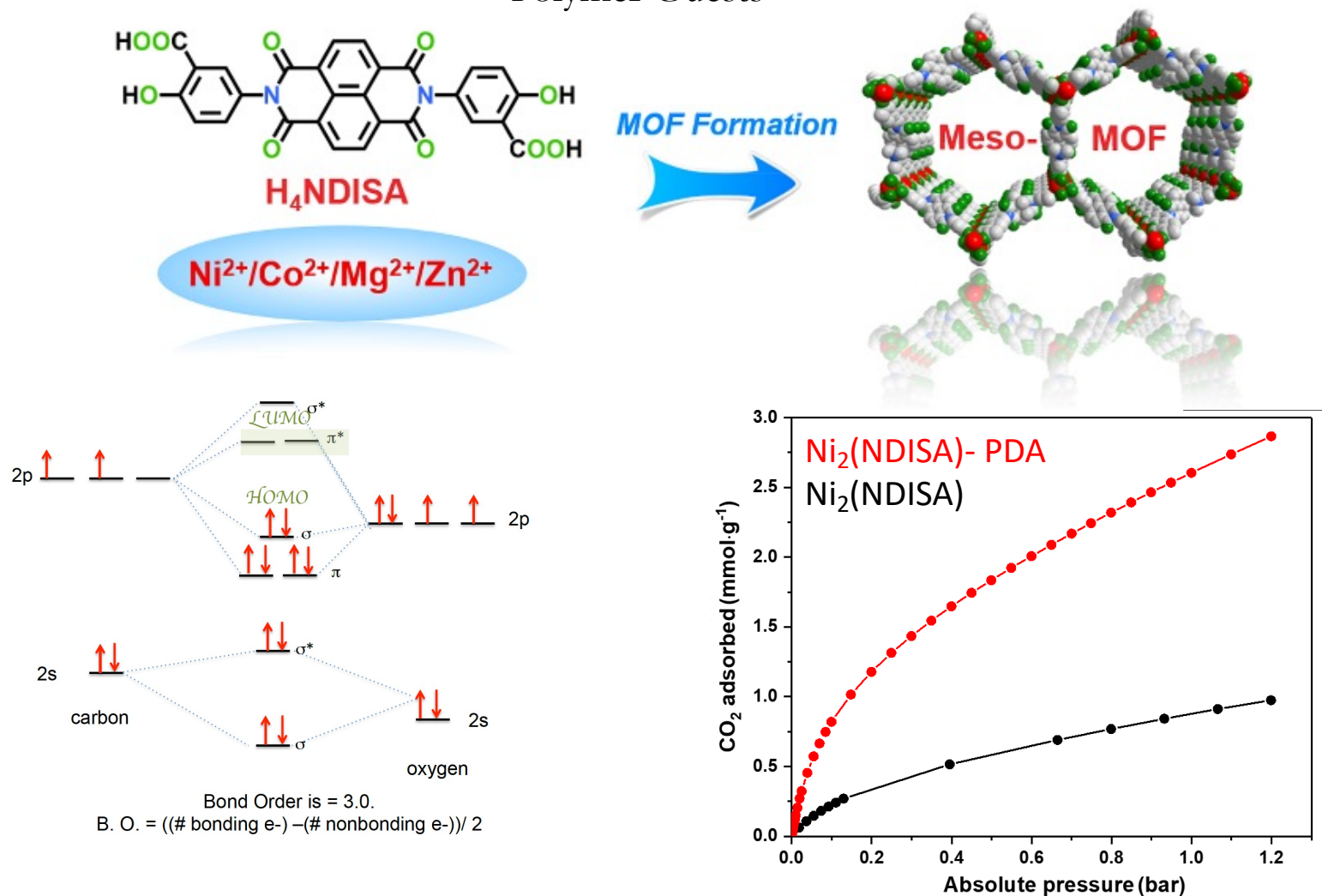
1579

1262

581

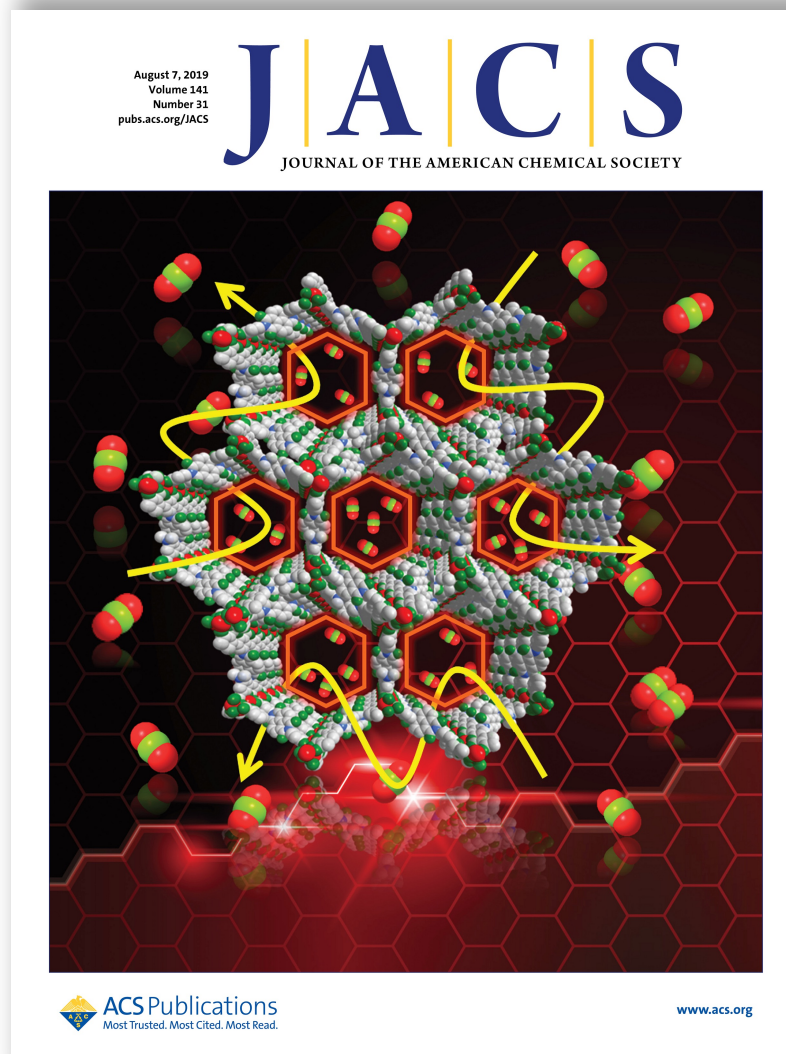
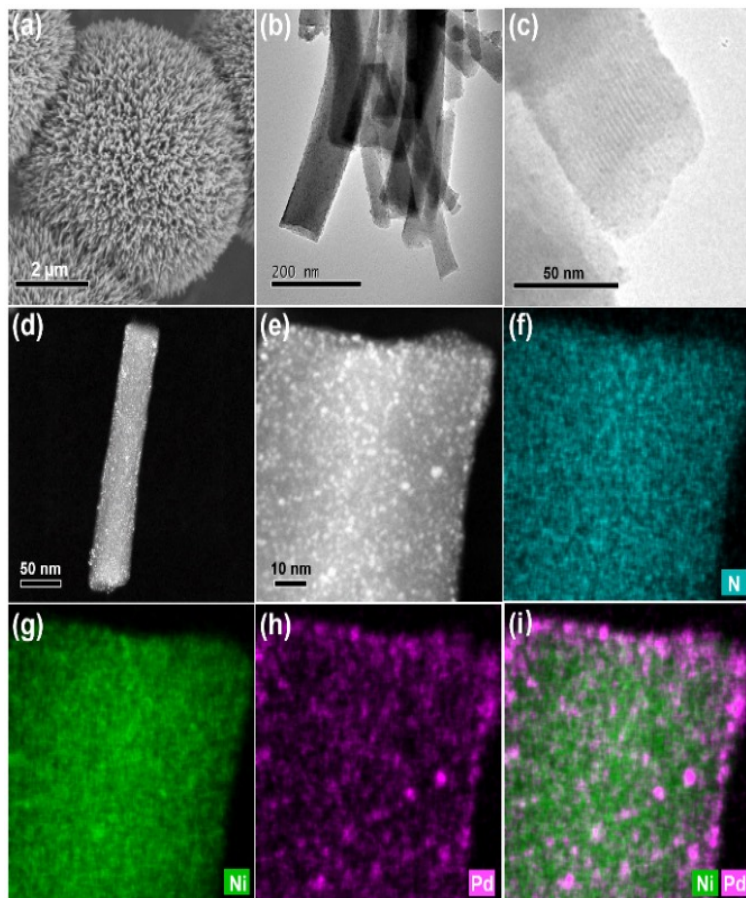
Achieving larger pore volumes and higher capacities

“Preserving Porosity of Mesoporous Metal-Organic Frameworks through the Introduction of Polymer Guests”



Achieving larger pore volumes and higher capacities

“Preserving Porosity of Mesoporous Metal-Organic Frameworks through the Introduction of Polymer Guests”



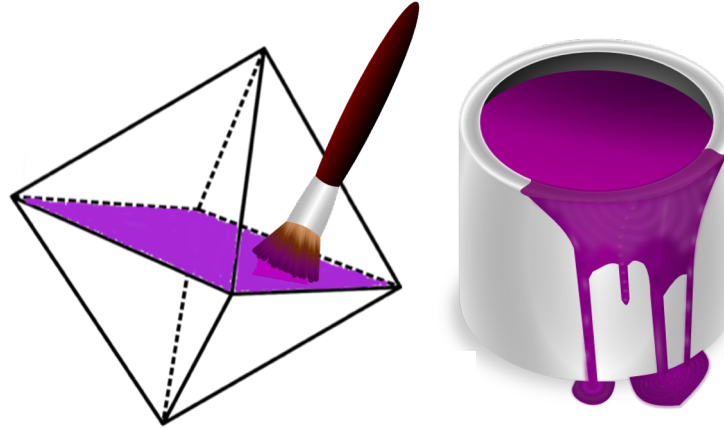
J. Am. Chem. Soc., 2019, 141, 12397.
Featured cover article.

End goals

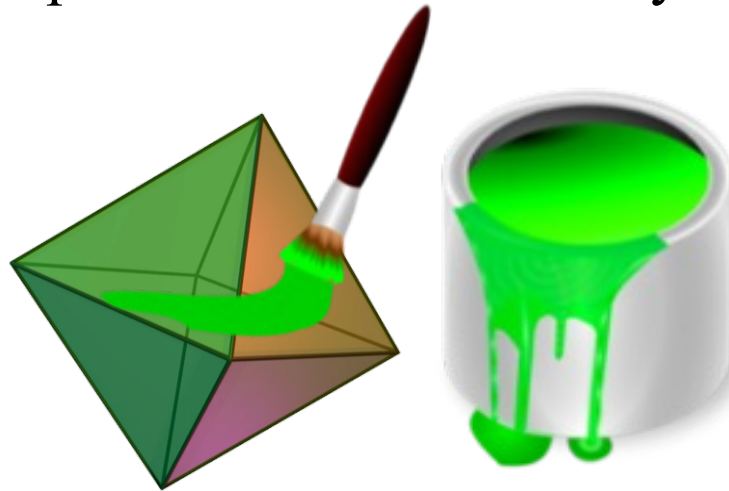
Objective 1: enhance adsorption/separation

Applications:

1. Heavy metal extraction
2. High value commodities
3. Harmful organics
4. Gas separations



Objective 2: enhance performance in humidity and acidic/basic media





**Laboratory of
Functional
Inorganic
Materials
(LFIM)**

Collaborators

Prof. Berend Smit, Sudi Jawahery, Mohamad Moosavi: EPFL

Prof. Michele Ceriotti, Rocio Semino: EPFL

Dr. Pascal Schouwink, Dr. Natalia Gasilova, Dr. Emad Oveisi: EPFL

Dr. Ralf Kägi: Eawag

Dr. Wouter Van Beek, Dr. Dimtry Chernishov: ESRF

Prof. Nazeeruddin, Dr. Aron Huckaba: EPFL

